ORIGINAL ARTICLE

A three-component coupling approach to the ACE-ring substructure of C19-diterpene alkaloids

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C19-diterpene alkaloids are a class of alkaloids with pharmacologically important activities having an intricately fused hexacyclic ABCDEF-ring system. Here we report expeditious assembly of the ACE-ring substructure 4a by applying a three-component coupling strategy. A radical-polar crossover reaction between an AE-ring radical precursor, a C-ring radical acceptor and an aldehyde was realized by the actions of Et_3B and O_2 , resulting in the installation of three new stereocenters and extension of the carbon chain corresponding to the B-ring. As the ACE-ring 4a possesses the correct C4,11-quaternary and C10-tertiary carbons, 4a would serve as an advanced intermediate for constructing the entire C19-diterpene alkaloid structures.

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INTRODUCTION

C19-diterpene alkaloids are a class of natural products present in the plants of the genera Aconitum and Delphinium, and many exhibit pharmacologically important biological activities. The structures of talatisamine and puberuline C are depicted in Scheme 1a as representative examples. The intricately fused hexacyclic ABCDEFring system of the C19-diterpene alkaloids has inspired chemists to invent synthetic methods for their assembly, 2-6 culminating in the full chemical construction of several members of this family.⁷⁻¹¹ Recently, we successfully synthesized the ABCDE-ring system¹² of talatisamine and the ABCDEF-ring system¹³ of puberuline C based on our development of a new radical-based strategy. In these synthetic studies, a C11-bridgehead radical of the AE-ring moiety undergoes cyclization with the C-ring enone to form a seven-membered B-ring. The success of the intramolecular C11-radical addition led us to explore its intermolecular version, because intermolecular multicomponent reactions generally ensure more convergent, and thus more efficient, approaches to complex molecular architectures. 14-19 Here we report the three-component coupling reaction between the bicyclic AE-ring, the 5-membered C-ring and the C6-8 carbon chain to assemble the ACE-ring substructure 4a with the correct C4,11-quaternary and C10-tertiary carbons in a single step (Scheme 1b).

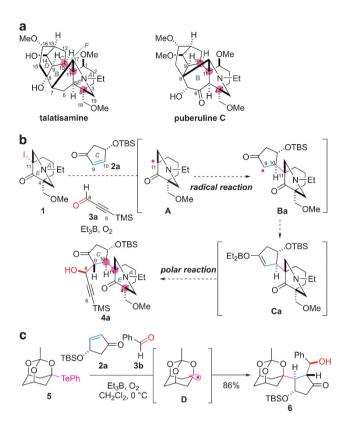
RESULTS AND DISCUSSION

We previously realized a three-component reaction between 2a, 3b and O,Te-acetal 5 using a reagent combination of Et_3B and O_2 (Scheme 1c). Parameter of O,Te-acetal 5 with Et_3B/O_2 generated the highly reactive bridgehead radical species D^{22-26} that sequentially coupled with α , β -unsaturated ketone 2a and aldehyde 3b via a radical-polar crossover mechanism to provide adduct 6. This method intermolecularly connected the three simple units with stereoselective

installation of the three new stereocenters, and thus significantly increased the molecular complexity in a single step.

To explore efficient strategies for synthesizing the C19-diterpene alkaloids, we decided to construct the ACE-ring system **4a** with the C6-8 carbon chain by employing the radical–polar three-component reaction (Scheme 1b). We planned to assemble the structure of **4a** from the azabicyclo[3.3.1]nonane AE-ring **1**, 5-membered C-ring **2a** and the C6-8 carbon chain **3a**. Et₃B/O₂-promoted C-I bond cleavage of **1** would produce the nucleophilic C11-bridgehead radical **A** that would add to the electron-deficient double bond of C-ring **2a**, leading to **Ba**. ^{27,28} Then, radical **Ba** would be captured by Et₃B to form boron enolate **Ca**²⁹ that would undergo the aldol reaction with aldehyde **3a**. ^{30–32} Hence, the one-pot radical and polar additions were expected to afford **4a** possessing the correct C4,11-quaternary and C10-tertiary carbon centers of the C19-diterpene alkaloids, such as talatisamine and puberuline C (highlighted by small circles).

Before investigating the key coupling reactions, we prepared the optically active AE-ring 1 from 2-(methoxycarbonyl)cyclohexanone (7) (Scheme 2). Bromination of 7, followed by exchange of the bromide of 8 with iodide using NaI, led to 9. The resultant 9 underwent the double Mannich reaction in the presence of formaldehyde and ethyl amine, giving rise to the azabicyclo[3.3.1]nonane structure (\pm) -10 as the racemate.³³ The methyl ester and the ketone groups of (\pm) -10 were in turn simultaneously reduced with DIBAL-H (diisobutylaluminium hydride) to the primary and secondary hydroxy groups of (\pm) -11 (dr at C5 = 1:1). Then, the racemic (\pm) -11 was subjected to enzymatic resolution to obtain enantio-enriched (-)-11. Although (\pm) -11 possesses two potentially reactive hydroxy groups for the enzymatic acylation, screening of enzymes and acylating reagents permitted us to realize the chemo- and enantioselective functionalization of the C18-primary hydroxy group. Namely,



Scheme 1 (a) Structures of the C19-diterpene alkaloids, talatisamine and puberuline C. (b) Plan for assembly of the AE-ring, C-ring and C6-8 carbon chain of the C19-diterpene alkaloids by the three-component coupling reaction. (c) Previously developed radical–polar crossover three-component coupling reaction.

treatment of diol (±)-11 (dr at C5=1:1) with Candida rugosa lipase $^{34-36}$ and vinyl butyrate in i-Pr₂O at 28 °C provided (–)-11 (39% yield) along with (+)-12 having a C18-butyrate group (39% yield). The C4,11-absolute configurations of (-)-11 were elucidated by NMR experiments of derivatives of 4a (Table 1, see Supplementary Information for details). To determine the enantiomeric ratio of the C5-diastereomeric mixtures (-)-11 and (+)-12, the corresponding C5-ketone 13a/b was prepared separately from these compounds, and then analyzed with the chiral HPLC. Acetylation of the C18-hydroxy group of (-)-11, and subsequent Dess-Martin oxidation of the C5-hydroxy group provided 13a and 13b in a 9:1 enantiomeric ratio. On the other hand, basic hydrolysis of the butyrate of (+)-12 was followed by acetylation and oxidation to afford 13a and 13b in a 1:5.5 ratio. Finally, compound (-)-11 (er = 9:1) was converted to the requisite AE-ring fragment 1 in two steps; site-selective methyl ether formation by the action of Me₃O•BF₄ and 2,6-di-tert-butylpyridine, and 2-azaadamantane N-oxyl (AZADO)-catalyzed C5-oxidation in the presence of CuCl.37

To evaluate the reactivity of AE-ring iodide 1 as a radical precursor, we first examined the formation of the corresponding radical $\bf A$ and subsequent addition to the cyclopentenone derivatives (2a–c) (Table 1). Upon treatment of 1 and cyclopentenone 2b with Et₃B and O₂ in CH₂Cl₂ at room temperature, the C-I bond at the congested bridgehead position was homolytically cleaved to generate bridgehead radical $\bf A$ that reacted with 2b to furnish 14b in 65% yield (dr at C10=1:1, entry 1). When the enantiopure cyclopentenone derivatives 2a and 2c³⁸ were used under the same conditions (entries 2 and 3), the

Scheme 2 Synthesis of optically active AE-ring fragment 1. Reagents and conditions: (a) $\rm Br_2,\ Et_2O;\ (b)\ Nal,\ acetone;\ (c)\ aq\ HCHO,\ aq\ EtNH_2,\ MeOH,\ 45\ ^{\circ}C,\ 83\%\ (3\ steps);\ (d)\ DIBAL-H,\ THF,\ 87\%,\ (dr\ at\ C5=1:1);\ (e)\ lipase from Candida rugosa,\ vinyl\ butyrate,\ {\it i-Pr}_2O,\ 28\ ^{\circ}C,\ 39\%\ for\ (-)-11\ (dr\ at\ C5=1:1),\ 39\%\ for\ (+)-12\ (dr\ at\ C5=1.7:1);\ (f)\ Me_3O•BF_4,\ 2,6-di-tert-butylpyridine,\ CH_2Cl_2;\ (g)\ AZADO,\ DMAP,\ 2,2'-bipyridine,\ CuCl,\ air,\ CH_3CN,\ 48\%\ (2\ steps);\ (h)\ Ac_2O,\ pyridine,\ 100\%\ from\ (-)-11;\ (i)\ Dess-Martin\ reagent,\ CH_2Cl_2,\ 80\%,\ 13a:\ 13b=9:\ 1;\ (j)\ 1M\ KOH,\ MeOH,\ 73\%\ from\ (+)-12;\ (k)\ Ac_2O,\ pyridine,\ 100\%;\ (l)\ Dess-Martin\ reagent,\ CH_2Cl_2,\ 84\%,\ 13a:\ 13b=1:\ 5.5.$

radical reaction of 1 occurred from the opposite side of the preexisting *tert*-butyldimethylsilyl (TBS)-oxy and acetonide groups, providing 14a (51%) and 14c (52%), respectively, in a C11-stereospecific and C10-stereoselective manner. Thus, the sterically cumbersome bond between the C11-quaternary and C10-tertiary carbon atoms of 14a-c were intermolecularly connected under mild conditions, corroborating the potent reactivity of radical A.

The bridgehead radical reaction was next extended to the radicalpolar crossover three-component reactions. Et₃B/O₂ successfully promoted the reaction between iodide 1, chiral cyclopentenone 2a and benzaldehyde (3b) in CH₂Cl₂ at room temperature to afford adduct 4b in a C8,9,10-stereoselective manner (53%, entry 4). (Trimethylsilyl)propynal (3a) participated in the coupling with 1 and 2a in the presence of Et₃B and O₂ to selectively yield 4a (55%, dr at C8 = 4:1, entry 5). No radical reaction to the triple bond of 3a or 4a was observed, showing the high chemoselectivity of the present method. Moreover, acid/base-sensitive aldehyde 3c with the β-silvloxy group functioned as an effective electrophile to provide 4c (52%, dr at C8 = 2.9:1, entry 6). The newly generated C8,9,10stereochemistry of coupling products 4a-c was consistent, and determined by extensive NMR experiments using their derivatives (see Supplementary Information for details). Significantly, the C9-substituted ACE-ring systems 4a-c with the two quaternary carbons (C4,11) and one tertiary carbon (C10) of the C19-diterpene alkaloids were built in a single operation in neutral media at room temperature. In contrast to the successful formation of 4a-c, the three-

Scheme 3 Rationale of the different reactivity between 2a and 2c, and the stereoselectivity of the three-component coupling reaction.

component adduct **4d** was obtained from **1**, **2c** and **3b** in only 10% yield, and the two-component adduct **14c** was mainly generated in 36% yield (entry 7). The distinct behavior of **2a** and **2c** indicated that the C-ring structure influenced the efficiency of the aldol reaction.

The reaction mechanism of the selective generation of 4b and 14c from 2a and 2c, respectively, is outlined in Scheme 3. Addition of bridgehead radical A to 2a from the opposite side of the TBS-oxy group installs the C10-stereochemistry. After the formation of boron enolate Ca, aldehyde 3b approaches from the less hindered face to avoid the bulky AE-ring, and forms the boron-chelating 6-membered transition state Ea, from which the C8,9-stereochemistry of 4b is established.^{39,40} In the case of 2c, the radical reaction of 2c and the radical termination of Et₃B occurred similarly to 2a, producing Cc. The approach of 3b toward one face of the enolate Cc, however, is blocked by the AE-ring, and the approach toward the other face is hindered by the acetonide-protected 1,2-diol. Because of the structural differences in the Y,Z-functionalities between Ec and Ea, the aldol reaction via Ec becomes less efficient compared with Ea. As a result, the yield of the three-component coupling adduct 4d is significantly decreased, and hydrolysis of Cc mainly occurs to produce the two-component adduct 14c.

In conclusion, we investigated the two- and three-component reactions of 1, and realized the expeditious assembly of the functionalized ACE-ring substructure 4a of the C19-diterpene alkaloids by applying the radical–polar crossover reaction. AE-ring 1, C-ring 2a and C6-8 chain 3a were coupled to generate 4a using Et₃B and O₂ through formation of the bridgehead radical from iodide 1, radical addition to cyclopentenone 2a and polar addition of the resultant boron enolate to aldehyde 3a. Remarkably, this operationally simple reaction enabled connection of the hindered C10-11 and C8-9 bonds, and installation of the C8,9,10-stereocenters under mild conditions in a single step. As the thus obtained 4a bears the C4,11-quaternary and C10-tertiary carbon centers, and the C6-8 chain of the C19-diterpene alkaloids, the compound would function as a valuable advanced intermediate for their total syntheses.

Table 1 The two- and three-component radical coupling reactions of $\mathbf{1}^a$

Entry	Cyclopentenone	Aldehyde	Results
1	0= <u></u>	-	O H 11 N Et OMe 14b: 65% (dr at C10 = 1 : 1)
2	O OTBS	-	OOTBS OOTBS OOTBS OOTBS OOTBS
3	$0 = \bigcup_{i=1}^{\infty} \bigcup_{i=1}^{\infty} 0$	-	O HIN Et
4	O CTBS	O Ph	14c: 52%b O 0 10 N Et Ph OMe 4b: 53%°
5	O OTBS	TMS	O O THE OTHER OTHE
6	O OTBS	OTBS 3c	OTBS OTBS OMe 4c: 52% (dr at C8 = 2.9:1)°
7		O Ph	O O O O O O O O O O O O O O O O O O O

^aConditions: **1** (er=9:1, 1 equiv), **2a-c** (3 equiv), **3a-c** (0 or 3 equiv), Et₃B (3 equiv), CH_2CI_2 , O_2 , room temperature.

b The product contained a small amount of the diastereomer that would be derived from the minor enantiomer of **1** (**14a**: its diastereomer = 9.1:1, **14c**: its diastereomer = 14:1). The product did not contain the diastereomer derived from the minor enantiomer of **1**. The product did not contain the diastereomer derived from the minor enantiomer of **1**. The variety of the product did not contain the diastereomer derived from the minor enantiomer of **1**. The variety of the product did not contain the diastereomer derived from the minor enantiomer of **1**.

d14c was generated in 36% yield.
 eThe yield was calculated by ¹H NMR analysis, because 4d and 14c were inseparable.

EXPERIMENTAL PROCEDURES

General methods

All reactions sensitive to air or moisture were carried out under argon atmosphere in dry solvents under anhydrous conditions, unless otherwise noted. THF, CH2Cl2 and Et2O were purified by Glass Contour solvent dispensing system (Nikko Hansen, Osaka, Japan). All other reagents were used as supplied unless otherwise stated. Analytical TLC was performed using E. Merck Silica gel 60 F254 precoated plates (Merck, Darmstadt, Germany). Preparative TLC was performed using E. Merck Silica gel 60 F254 precoated plates with 0.50 mm thickness. Flash chromatography was performed using 40-50 μm Silica Gel 60N (Kanto Chemical, Tokyo, Japan), 40-100 μm Silica Gel 60N (Kanto Chemical), 100-210 µm Silica Gel 60N (Kanto Chemical) and 32-53 µm Silica gel BW-300 (Fuji Silysia Chemical, Aichi, Japan). Optical rotations were measured on a JACSO P-200 Digital Polarimeter at room temperature using the sodium D line (JASCO, Tokyo, Japan). IR spectra were recorded on a JASCO FT/IR-4100 spectrometer. ¹H- and ¹³C-NMR spectra were recorded on a JEOL JNM-ECX-500 (500 MHz), a JNM-ECA-500 (500 MHz) or a JNM-ECS-400 (400 MHz) spectrometer (JEOL, Tokyo, Japan). Chemical shifts were reported in ppm on the δ scale relative to CHCl₃ (δ = 7.26 for ¹H NMR), CDCl₃ ($\delta = 77.0$ for ¹³C NMR), C₆D₅H ($\delta = 7.16$ for ¹H-NMR) and C_6D_6 ($\delta = 128.0$ for ¹³C-NMR) as internal references. Signal patterns are indicated as s, singlet; d, doublet; t, triplet; q, quartet; m, multiplet; br, broad peak. The numbering of compounds corresponds to that of natural product. Electrospray ionization mass spectra were measured on a JEOL JMS-T100LP or a Bruker microTOF II instrument.

Iodide (+)-10

Br₂ (200 µl, 7.96 mmol) was added to a solution of 7 (1.13 g, 7.24 mmol) in Et₂O (4.8 ml) at 0 °C. The reaction mixture was warmed to room temperature and stirred for 3 h. Then, saturated aqueous NaHCO₃ (5 ml) was added. The resulting mixture was extracted with Et₂O (5 ml \times 3), and the combined organic layers were washed with brine (10 ml), dried over Na₂SO₄, filtered and concentrated to afford the crude bromide 8. The crude 8 was divided into three equal parts. One-third of the crude 8 was used in the next iodination.

NaI (470 mg, 3.13 mmol) was added to a solution of one-third of the above crude bromide 8 (2.41 mmol) in acetone (4.8 ml) at room temperature. The reaction mixture was stirred at room temperature for 1.5 h. Then, the reaction mixture was filtered through a pad of Celite and the filter cake was washed with acetone. The resultant solution was concentrated to afford the crude 9 that was used in the next reaction without further purification.

Aqueous EtNH₂ (70% in water, 465 μl, 7.23 mmol) was added to a solution of the above crude 9 in MeOH (8.0 ml) and aqueous HCHO (37% in water, 2.3 ml, 28.9 mmol) at 0 °C over 3 h. The reaction mixture was warmed to 45 °C and stirred for 5 h. After the mixture was cooled to room temperature, H₂O (16 ml) was added. The resultant mixture was extracted with EtOAc (8 ml ×3), and the combined organic layers were washed with brine (5 ml), dried over Na₂SO₄, filtered and concentrated. The residue was purified by flash column chromatography on silica gel (100 g, hexane/EtOAc 50:1 to 5:1) to afford iodide (±)-10 (703 mg, 2.00 mmol). The yield was determined to be 83% yield over 3 steps based on one-third amount of the starting compound 7: pale yellow oil; IR (film) v 2969, 2949, 2932, 2811, 1739, 1731, 1454, 1435, 1292, 1259, 1224, 1208, 1175, 1131, 1112, 1098, 1090 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 1.11 (3H, t, J=7.3 Hz, NCH₂CH₃), 1.51 (1H, m), 2.30 (1H, dddd, J = 14.2, 6.4, 2.3, 2.3 Hz), 2.43 (2H, m, NC H_2 CH₃), 2.57 (1H, m), 2.80 (1H, m), 2.99 (1H, dddd, J=13.7, 5.9, 2.3, 2.3 Hz), 3.03 (1H, dd, J=11.9, 1.9)2.3 Hz, NCH_AH_B), 3.16 (1H, m, NCH_AH_B), 3.18 (1H, m), 3.30 (1H, dd, J=11.9, 2.3 Hz, NCH_AH_B), 3.74 (1H, dd, J=8.7, 2.8 Hz), 3.76 (3H, s, OMe); ¹³C NMR (100 MHz, CDCl₃) δ 12.6, 24.4, 36.6, 49.4, 50.3, 52.5, 56.2, 59.0, 61.2, 71.6, 170.5, 203.1; HRMS (ESI) calcd for C₁₂H₁₈INO₃Na [M+Na]⁺ 374.0224, found 374.0225.

Diol (±)-11

DIBAL-H (1.0 $\,\mathrm{M}$ in hexane, 9.2 ml, 9.2 mmol) was added to a solution of (\pm)-10 (536 mg, 1.53 mmol) in THF (15 ml) at 0 °C. The reaction mixture was warmed to room temperature and stirred for 30 min. Then, saturated aqueous NH₄Cl (10 ml), saturated aqueous potassium sodium tartrate (15 ml)

and EtOAc (15 ml) were successively added. After being stirred at room temperature for 1 h, the resultant mixture was extracted with EtOAc (10 ml × 3). The combined organic layers were dried over Na₂SO₄, filtered and concentrated. The residue was purified by flash column chromatography on silica gel (40 g, hexane/EtOAc 50:1 to 5:1) to afford a 1:1 C5-diastereomeric mixture of diol (±)-11 (431 mg, 1.33 mmol) in 87% yield: yellow oil; IR (film) ν 3403, 2970, 2925, 2809, 1471, 1452, 1394, 1327, 1291, 1227, 1146, 1076, 1025, 959, 943 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 1.03 (3H × 1/2, t, J = 6.8 Hz, N (CH_2CH_3) , 1.04 (3H ×1/2, t, J=7.2 Hz, $N(CH_2CH_3)$, 1.32–1.50 (2H and $1H \times 1/2$, m), 1.44-1.48 ($1H \times 1/2$, m), 2.03-2.10 (1H, m), 2.23-2.40 (3H, m), 2.54 (1H × 1/2, d, J = 9.2 Hz), 2.58–2.93 (5H and 1H × 1/2, m), 2.97 (1H × 1/2, dd, $J = 8.7 \cdot 1.9 \text{ Hz}$), 3.18 (1H, m), 3.34 (1H, m), 3.43 (1H × 1/2, d, J = 8.7 Hz), 3.48 (1H \times 1/2, d, J=9.2 Hz), 3.59 (1H \times 1/2, dd, J=9.2, 1.8 Hz), 3.88 $(1H \times 1/2, s)$, 3.89 $(1H \times 1/2, s)$; ¹³C NMR $(100 \text{ MHz}, \text{CDCl}_3)$ δ 12.5 (1C $\times 1/2$), 12.6 (1C $\times 1/2$), 24.8 (1C $\times 1/2$), 25.3 (1C $\times 1/2$), 25.4 (1C $\times 1/2$), 33.9 (1C), 38.9 (1C \times 1/2), 42.3 (1C \times 1/2), 45.9 (1C \times 1/2), 51.2 (1C \times 1/2), 51.7 $(1C \times 1/2)$, 53.0 $(1C \times 1/2)$, 60.4 $(1C \times 1/2)$, 62.5 $(1C \times 1/2)$, 63.4 $(1C \times 1/2)$, 69.0 (1C \times 1/2), 70.3 (1C \times 1/2), 70.9 (1C \times 1/2), 80.3 (1C), 81.1 (1C \times 1/2); HRMS (ESI) calcd for C₁₁H₂₁INO₂ [M+H]⁺ 326.0611, found 326.0622.

Diol (-)-11 and butyrate (+)-12

Lipase from C. rugosa (1009 U mg⁻¹, 1.27 g) and vinyl butyrate (1.2 ml, 9.7 mmol) were successively added to a solution of diol (\pm)-11 (1.27 g, 3.91 mmol) in i-Pr2O (40 ml) at 28 °C. The reaction mixture was stirred at 28 °C for 1 h. Then, the mixture was filtered through a pad of Celite with Et₂O. After the filtrate was concentrated, brine (15 ml) was added. The resultant solution was extracted with EtOAc (15 ml × 3), and the combined organic layers were dried over Na2SO4, filtered and concentrated. The residue was purified by flash column chromatography on silica gel (30 g, hexane/EtOAc 50:1 to 1:1) to afford a 1:1 C5-diastereomeric mixture of diol (-)-11 (491 mg, 1.51 mmol) and a 1.7:1 C5-diastereomeric mixture of butyrate (+)-12 (632 mg, 1.54 mmol) in 39% and 39% yields, respectively. The enantiopurity of (-)-11 and (+)-12 was evaluated by the chiral HPLC analysis of compound 13 derived from (-)-11 and (+)-12, respectively (see Scheme 2). The enantiomeric ratio (er) of 13a/13b from (–)-11 was 9:1 and that from (+)-12 was 1:5.5. The analytical detail was described in the Supplementary Information. Diol (–)-11: yellow oil; $[\alpha]_D^{23}$ -3.5 (c 1.00, CHCl₃). Butyrate (+)-12: $[\alpha]_D^{26}$ 8.0 (c 1.00, $CHCl_3$); IR (film) ν 3500, 2966, 2931, 2875, 2807, 1739, 1454, 1416, 1384, 1304, 1261, 1181, 1132, 1081, 1048, 1091, 991 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.95 (3H \times 5/8, t, J=7.4 Hz, COCH₂CH₂CH₃), 1.03 (3H \times 3/8, t, J=7.3 Hz, NCH_2CH_3), 1.33 (1H × 3/8, m), 1.42-1.46 (2H × 5/8, m), 1.67 (2H, qt, J = 7.4, 7.4 Hz, $COCH_2CH_2CH_3$), 1.77 (1H, m), 2.19–2.33 (5H, m), 2.40 (1H × 3/8, d, I = 11.0 Hz, NH_AH_B), 2.55 (1H × 5/8, d, I = 11.0 Hz, NH_AH_B), 2.56–2.70 (1H, m), 2.72–2.90 (3H, m), 2.95 (1H \times 5/8, d, J=11.0 Hz, NH_AH_B), 3.19 (1H \times 5/8, s, H-5), 3.56 (1H \times 5/8, d, J = 11.0 Hz, NH_AH_B), 3.74-3.80 (1H and 1H \times 3/8, m), 4.06 (1H \times 5/8, d, J=11.0 Hz, H-18a), 4.10 (1H \times 3/8, d, J = 11.0 Hz, H-18a); ¹³C NMR (125 MHz, C₆D₆) δ 12.5, 12.6, 13.77, 13.79, 18.7, 18.8, 25.5, 25.7, 25.8, 33.8, 36.1, 36.2, 39.4, 41.9, 42.1, 46.6, 51.4, 51.8, 53.6, 60.8, 61.8, 63.0, 63.1, 69.36, 69.39, 69.9, 76.9, 78.2, 173.1, 173.2; HRMS (ESI) calcd for C₁₅H₂₇INO₃ [M+H]⁺ 396.1030, found 396.1017.

Iodide 1

Me $_3$ O•BF $_4$ (564 mg, 3.81 mmol) was added to a solution of a 1:1 C5-diastereomeric mixture of diol ($^-$)-11 (620 mg, 1.91 mmol) and 2,6-ditert-butylpyridine (1.3 ml, 5.7 mmol) in CH $_2$ Cl $_2$ (38 ml) at 0 °C. The reaction mixture was warmed to room temperature and stirred for 2 h. Then, saturated aqueous NaHCO $_3$ (20 ml) was added. The resultant mixture was extracted with EtOAc (15 ml \times 3), and the combined organic layers were dried over Na $_2$ SO $_4$, filtered and concentrated. The residue was purified by flash column chromatography on silica gel (30 g, hexane/EtOAc 50:1 to 5:1) to afford the crude methyl ether that was used in the next reaction without further purification.

CuCl (227 mg, 2.29 mmol) and AZADO (87.2 mg, 573 μ mol) were successively added to a solution of the above crude methyl ether, DMAP (93.3 mg, 764 μ mol) and 2,2′-bipyridine (59.7 mg, 382 μ mol) in CH₃CN (9.6 ml) at room temperature. The reaction mixture was stirred at room temperature for 15 min. Then, saturated aqueous Na₂S₂O₃ (10 ml) was added. The resultant

mixture was extracted with Et₂O (10 ml × 3), and the combined organic layers were dried over Na₂SO₄, filtered and concentrated. The residue was purified by flash column chromatography on silica gel (20 g, hexane/EtOAc 80:1 to 10:1) to afford iodide 1 (309 mg, 916 μmol) in 48% yield over 2 steps: colorless oil; [α] $_{\rm D}^{22}$ – 0.59 ($_{\rm c}$ 1.00, CHCl₃); IR (film) $_{\rm c}$ 2973, 2925, 2894, 2807, 1727, 1452, 1385, 1348, 1318, 1289, 1235, 1202, 1164, 1112, 1020, 965, 935 cm $_{\rm c}^{-1}$; $_{\rm c}^{1}$ H NMR (400 MHz, CDCl₃) δ 1.09 (3H, t, $_{\rm c}$ 1=7.3 Hz, NCH₂CH₃), 1.44 (1H, m), 1.75 (1H, ddd, $_{\rm c}$ 1=13.3, 13.3, 2.3 Hz), 2.34–2.40 (4H, m), 2.74 (1H, ddd, $_{\rm c}$ 1=13.3, 13.3, 2.3 Hz), 2.97 (1H, m), 3.09 (1H, d, $_{\rm c}$ 1=11.0 Hz), 3.18 (1H, m), 3.31–3.37 (2H, m), 3.34 (3H, s, OMe), 3.44 (1H, d, $_{\rm c}$ 1=1.0 Hz), 3.73 (1H, dd, $_{\rm c}$ 1=11.0, 2.3 Hz); $_{\rm c}$ 13°C NMR (100 MHz, CDCl₃) δ 12.6, 24.6, 37.5, 49.8, 50.3, 51.6, 59.2, 59.4, 61.9, 71.7, 76.2, 207.3; HRMS (ESI) calcd for C₁₂H₂₀INO₂ [M+Na] + 360.0431, found 360.0432.

General procedure A: two-component radical coupling reaction Compound 14b

Et₃B (0.99 M in hexane, 370 μl, 370 μmol) was added to a solution of cyclopentenone 2b (31 µl, 370 µmol) and iodide 1 (41.3 mg, 123 µmol) in CH₂Cl₂ (250 µl) at 0 °C over 30 min. The mixture was warmed to room temperature under air and stirred for 30 min. Then, saturated aqueous NaHCO₃ (2 ml) was added. The resultant mixture was extracted with EtOAc (2 ml × 3), and the combined organic layers were passed through a pad of silica gel with EtOAc. After the filtrate was concentrated, the residue was purified by flash column chromatography on silica gel (4 g, hexane/EtOAc 100:1 to 1:1) to afford a 1:1 C10-diastereomeric mixture of 14b (23.4 mg, 79.8 µmol) in 65% yield: $[\alpha]_D^{26}$ = 0.43 (c 1.00, CHCl₃); IR (film) ν 2957, 2930, 1731, 1710, 1459, 1362, 1330, 1208, 1165, 1112, 1058 cm $^{-1}$; ¹H NMR (500 MHz, CDCl₃) δ 1.08 $(3H \times 1/2, t, J = 6.9 \text{ Hz}, NCH_2CH_3), 1.09 (3H \times 1/2, t, J = 6.9 \text{ Hz}, NCH_2CH_3),$ 1.40-1.50 (1H, m), 1.63-1.70 (1H, m), 1.73-1.80 (2H, m), 1.95-2.07 (2H, m), 2.10-2.19 (2H, m), 2.21-2.27 (1H, m), 2.29-2.42 (7H, m), 2.72-2.90 (1H, m), 2.92 (1H \times 1/2, dd, J = 11.0, 1.8 Hz, N H_AH_B), 3.01 (1H \times 1/2, dd, J = 11.0, 1.8 Hz, NH_AH_B), 3.14 (1H × 1/2, dd, J = 11.5 Hz, NH_AH_B), 3.17 (1H × 1/2, dd, J = 11.5 Hz, NH_AH_B), 3.29 (1H × 1/2, d, J = 9.2 Hz, H-18a), 3.30 (1H × 1/2, d, J=9.2 Hz, H-18a), 3.34 (3H, s, OMe), 3.38 (1H, d, J=9.2 Hz, H-18b); ¹³C NMR (125 MHz, CDCl₃) δ 12.56, 12.57, 20.36, 20.37, 20.53, 20.54, 23.4, 23.7, 35.9, 36.3, 36.41, 36.42, 36.52, 36.53, 38.5, 38.7, 39.5, 39.8, 40.9, 41.2, 50.8, 50.9, 51.1, 52.3, 59.2, 59.3, 61.5, 61.6, 63.2, 63.6, 212.5, 212.6, 218.5, 218.7; HRMS (ESI) calcd for C₁₇H₂₇NO₃Na [M+Na]⁺ 316.1883, found 316.1880.

Compound 14a

According to the general procedure A, a 9.1:1 mixture of 14a and the diastereomer presumably originated from the minor enantiomer of 1 (25.7 mg, 60.7 µmol) was obtained in 51% yield by using cyclopentenone 2a (enantiopure, 75.8 mg, 359 μ mol), iodide 1 (er = 9:1, 40.1 mg, 119 μ mol) and Et₃B (0.99 M in hexane, 360 μl, 360 μmol) in CH₂Cl₂ (240 μl). The crude was purified by flash column chromatography on silica gel (4 g, CH₂Cl₂/EtOAc 100:1 to 1:1). $[\alpha]_D^{26}$ 23.0 (c 1.00, CHCl₃); IR (film) ν 2950, 2929, 2895, 2808, 1747, 1707, 1470, 1389, 1361, 1254, 1204, 1164, 1112, 1007, 979, 940, 908 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) peaks of the major isomer: δ 0.03 (3H, s, CH₃ of TBS), 0.07 (3H, s, CH₃ of TBS), 0.86 (9H, s, t-Bu of TBS), 1.09 (3H, t, J = 7.3 Hz, NCH₂CH₃), 1.46 (1H, m), 1.65–1.82 (2H, m), 2.15–2.28 (5H, m), 2.33 (1H, d, J=7.8 Hz), 2.36 (2H, q, J=7.3 Hz, NC H_2 CH₃), 2.58 (1H, d, J = 11 Hz, NCH_AH_B), 2.70 (1H, dd, J = 17.8, 7.4 Hz), 2.84 (1H, m), 2.95 (1H, dd, J = 11.0, 2.3 Hz, NCH_AH_B), 3.14 (1H, ddd, J = 11.4, 1.8, 1.8 Hz, NCH_AH_B), 3.32 (1H, d, J=9.6 Hz, H-18a), 3.33 (3H, s, OMe), 3.37 (1H, d, J = 9.6 Hz, H-18b), 4.50 (1H, dd, J = 6.0, 6.0 Hz, H-12); ¹³C NMR (100 MHz, CDCl₃) peaks of the major isomer: $\delta - 4.7$, - 3.9, 12.8, 17.7, 20.4, 25.7, 37.9, 40.3, 40.6, 49.3, 50.4, 50.8, 51.4, 52.0, 59.6, 62.3, 62.8, 70.9, 76.0, 216.1, 217.2; HRMS (ESI) calcd for C₂₃H₄₁NO₄SiNa [M+Na]⁺ 446.2697, found 446.2708.

Compound 14c

According to the general procedure A, a 14:1 mixture of **14c** and the diastereomer presumably originated from the minor enantiomer of **1** (23.4 mg, 64.0 μ mol) was obtained in 52% yield by using cyclopentenone **2c** (enantiopure, 56.8 mg, 368 μ mol), iodide **1** (er=9:1, 41.4 mg, 123 μ mol) and Et₃B (0.99 M in hexane, 370 μ l, 370 μ mol) in CH₂Cl₂ (250 μ l). The crude

was purified by flash column chromatography on silica gel (4 g, CH₂Cl₂ to CH₂Cl₂/EtOAc 10:1). $[\alpha]_D^{22}$ -65.0 (*c* 1.00, CHCl₃); IR (film) ν 2979, 2934, 2810, 1757, 1706, 1453, 1381, 1242, 1210, 1157, 1112, 1041, 1004, 976 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) peaks of the major isomer: δ 1.10 (3H, t, J=7.5 Hz, NCH₂CH₃), 1.33 (3H, s, acetonide), 1.43 (3H, s, acetonide), 1.50 (1H, m), 1.67 (1H, ddd, J=12.6, 12.6, 6.3 Hz), 1.81 (1H, ddd, J=13.2, 13.2, 6.3 Hz), 1.97 (1H, m), 2.04 (1H, m), 2.22 (1H, m), 2.38–2.45 (3H, m), 2.53 (1H, d, J=11.5 Hz, NCH_AH_B), 2.63 (1H, dd, J=19.5, 10.3 Hz), 2.67 (1H, m), 2.76 (1H, d, J=11.5 Hz, NCH_AH_B), 2.97 (1H, dd, J=11.5, 1.7 Hz, NCH_AH_B), 3.09 (1H, dd, J=11.5, 1.2 Hz, NCH_AH_B), 3.26 (1H, d, J=9.8 Hz, H-18a), 3.32 (3H, s, OMe), 3.36 (1H, d, J=9.8 Hz, H-18b), 4.61 (1H, d, J=6.9 Hz, H-12), 4.68 (1H, d, J=6.9 Hz, H-13); ¹³C NMR (100 MHz, CDCl₃) peaks of the major isomer: δ 12.6, 19.6, 24.7, 26.9, 37.2, 37.8, 38.9, 45.7, 50.2, 51.2, 51.7, 59.6, 62.2, 63.0, 75.8, 79.65, 79.68, 111.8, 212.8, 217.4; HRMS (ESI) calcd for C₂₀H₃₁NO₅Na [M+Na]⁺ 388.2094, found 388.2088.

General procedure B: three-component coupling reaction Compound 4b

Et₃B (0.99 M in hexane, 350 μl, 350 μmol) was added to a solution of cyclopentenone 2a (enantiopure, 75.4 mg, 355 μmol), aldehyde 3b (36 μl, 350 μ mol) and iodide 1 (er = 9:1, 39.9 mg, 118 μ mol) in CH₂Cl₂ (240 μ l) at 0 °C over 30 min. The reaction mixture was warmed to room temperature under air and stirred for 30 min. Then, saturated aqueous NaHCO₃ (2 ml) was added. The resultant mixture was extracted with EtOAc (2 ml × 3), and the combined organic layers were passed through a pad of silica gel with EtOAc. After the filtrate was concentrated, the residue was purified by flash column chromatography on silica gel (4 g, CH2Cl2/EtOAc 100:1 to 1:1) to afford 4b (33.2 mg, 62.7 μ mol) in 53% yield: colorless oil; [α] $_{\rm D}^{26}$ 3.13 (c 1.00, CHCl $_{\rm 3}$); IR (film) v 3462, 2952, 2928, 2894, 2857, 2807, 1708, 1472, 1456, 1388, 1253, 1172, 1113, 1061 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.11 (3H, s, CH₃ of TBS), 0.19 (3H, s, CH₃ of TBS), 0.94 (9H, s, t-Bu of TBS), 0.98 (3H, t, J = 6.9 Hz, NCH₂CH₃), 1.34-1.38 (1H, m), 1.44-1.53 (2H, m), 1.61 (1H, ddd, J = 12.0, 12.0, 6.3 Hz), 1.68 (1H, br d, J = 10.9 Hz), 2.12-2.19 (5H, m), 2.34 (1H, d, J = 18.9 Hz), 2.36 (1H, m), 2.64 (1H, dd, J = 11.4, 1.8 Hz), 2.64-2.74 (2H, m), 3.02 (1H, dd, J=11.4, 1.4 Hz), 3.13 (1H, d, J=9.6 Hz, H-18a), 3.33(3H, s, OMe), 3.39 (1H, d, J = 9.6 Hz, H-18b), 4.43 (1H, d, J = 6.0 Hz), 4.48 (1H, s), 4.92 (1H, d, J = 7.8 Hz), 7.29-7.38 (5H, m, aromatic); ¹³C NMR $(100 \text{ MHz}, \text{CDCl}_3) \delta - 4.6, 12.5, 17.8, 20.0, 25.7, 37.3, 37.4, 49.6, 50.3, 51.2,$ 52.1, 55.3, 56.7, 59.5, 62.1, 63.1, 71.1, 76.0, 76.3, 127.2, 127.8, 128.2, 141.2, 216.4, 220.4; HRMS (ESI) calcd for C₃₀H₄₇NO₅SiNa [M+Na]⁺ 552.3116, found 552.3117.

Compound 4a

According to the general procedure B, C8(S)-4a (29.0 mg, 52.7 µmol) and C8 (R)-4a (7.4 mg, 13.5 μ mol) were obtained in 44% and 11% yields, respectively, by using cyclopentenone 2a (enantiopure, 76.7 mg, 361 µmol), aldehyde 3a $(53 \mu l, 361 \mu mol)$, iodide 1 (er = 9:1, 40.6 mg, 120 μ mol) and Et₃B (0.99 M in hexane, 370 μ l, 370 μ mol) in CH₂Cl₂ (240 μ l). The crude was purified by flash column chromatography on silica gel (4 g, CH₂Cl₂/EtOAc 100:1 to 5:1). C8(S)-**4a**: $[\alpha]_D^{26}$ **–** 0.11 (*c* 1.00, CHCl₃); IR (film) ν 3454, 2955, 2929, 2897, 2857, 2809, 2175, 1743, 1709, 1472, 1462, 1388, 1361, 1286, 1251, 1172, 1112, 1066, 1006, 976 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.07 (3H, s, CH₃ of TBS), 0.13 (3H, s, CH₃ of TBS), 0.16 (9H, s, CH₃ of TMS), 0.86 (9H, s, t-Bu of TBS), 1.09 (3H, t, J=6.8 Hz, NCH₂CH₃), 1.49 (1H, m), 1.66 (1H, m), 1.87 (1H, m), 2.02 (1H, m), 2.22-2.41 (7H, m), 2.54 (1H, d, J = 11.0 Hz, NCH_AH_B), 2.81 <math>(2H, dd, J = 11.0 Hz, NCH_AH_B)J = 18.3, 5.5 Hz), 3.01 (1H, d, J = 11.0 Hz, NCH_AH_B), 3.13 (1H, d, J = 11.0 Hz, NCH_AH_B), 3.26 (1H, d, J=9.6 Hz, H-18a), 3.32 (3H, s, OMe), 3.36 (1H, d, J = 9.2 Hz, H-18b), 3.70 (1H, s, OH), 4.43 (1H, d, J = 5.5 Hz, H-12), 4.66 (1H, d, J=7.8 Hz, H-8); ¹³C NMR (100 MHz, CDCl₃) $\delta -4.7$, -4.5, -0.2, 12.7, 17.7, 20.0, 25.6, 37.4, 38.9, 49.1, 50.4, 51.3, 52.3, 55.3, 55.5, 59.6, 62.2, 63.8, 65.3, 71.0, 75.9, 91.0, 103.6, 216.7, 218.6; HRMS (ESI) calcd for C₂₉H₅₁NO₅-Si₂Na [M+Na]⁺ 572.3198, found 572.3192. C8(R)-4a: $[\alpha]_D^{25}$ 5.21 (c 1.00, CHCl₃); IR (film) ν 3441, 2955, 2929, 2898, 2857, 2810, 2172, 1743, 1713, 1471, 1459, 1389, 1250, 1173, 1113, 1065 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.07 (3H, s, CH₃ of TBS), 0.14 (3H, s, CH₃ of TBS), 0.16 (9H, s, CH₃ of TMS), 0.87 (9H, s, t-Bu of TBS), 1.08 (3H, t, J = 7.3 Hz, NCH₂CH₃), 1.25 (1H, s, OH), 1.45, (1H, m), 1.71 (1H, m), 1.94 (1H, m), 2.13 (1H, m), 2.23-2.44 (7H, m), 2.49 (1H, d, J=11.0 Hz, NC H_A H $_B$), 2.69 (1H, m), 2.80 (1H, dd, J=12.0, 6.4 Hz), 2.88 (1H, dd, J=11.0, 1.8 Hz, NC H_A H $_B$), 3.12 (1H, dd, J=11.0, 1.4 Hz, NC H_A H $_B$), 3.29 (1H, d, J=9.6 Hz, H-18a), 3.32 (3H, s, OMe), 3.34 (1H, d, J=9.6 Hz, H-18b), 4.39 (1H, d, J=4.4 Hz, H-12), 4.70 (1H, d, J=7.8 Hz, H-8); ¹³C NMR (125 MHz, CDCl $_3$) δ -4.8, -4.5, -0.2, 12.6, 17.7, 20.4, 25.7, 37.4, 38.5, 49.0, 50.3, 51.3, 52.4, 55.2, 55.3, 59.6, 62.0, 63.6, 65.2, 70.8, 75.8, 77.3, 103.6, 217.0, 218.4, one ¹³C peak was not observed; HRMS (ESI) calcd for C $_{29}$ H $_{51}$ NO $_{5}$ Si $_{2}$ Na [M+Na] $^+$ 572.3198, found 572.3200.

Compound 4c

According to the general procedure B, a 2.9:1 C8-diastereomeric mixture of 4c (39.2 mg, 64.0 µmol) was obtained in 52% yield by using cyclopentenone 2a (enantiopure, 78.4 mg, 369 μmol), aldehyde 3c (69.5 mg, 369 μ mol), iodide 1 (er = 9:1, 41.5 mg, 123 μ mol) and Et₃B (0.99 μ in hexane, 380 μ l, 380 μ mol) in CH₂Cl₂ (250 μ l). The crude was purified by flash column chromatography on silica gel (4 g, CH₂Cl₂/EtOAc 100:1 to 5:1). $[\alpha]_D^{26}$ 2.48 (c 1.00, CHCl₃); IR (film) ν 3471, 2953, 2930, 2892, 2857, 2810, 1740, 1470, 1387, 1362, 1295, 1254, 1188, 1171, 1006, 973, 938 cm $^{-1};$ $^{1}\mathrm{H}$ NMR (400 MHz, CDCl₃) peaks of the major isomer: δ 0.04 (6H, s, CH₃ of TBS \times 2), 0.06 (3H, s, CH_3 of TBS), 0.11 (3H, s, CH_3 of TBS), 0.87 (18H, s, t-Bu of TBS \times 2), 1.06 (3H, t, J = 7.3 Hz, NCH₂CH₃), 1.42–1.44 (1H, m), 1.60–1.80 (3H, m), 1.93 (1H, m), 2.02-2.06 (1H, m), 2.13 (1H, s), 2.20-2.37 (6H, m), 2.47 (1H, d, $J=11.0 \text{ Hz}, \text{ NC}H_AH_B), 2.67-2.73 (1H, m), 2.82 (1H, m), 2.96 (1H, d,$ J=11.0 Hz, NCH_A H_B), 3.14 (1H, dd, J=11.4, 1.8 Hz, NCH_A H_B), 3.26 (1H, d, J = 10.1 Hz, H-18a), 3.31 (3H, s, OMe), 3.35 (1H, d, J = 9.6 Hz, H-18b), 3.80 (2H, m), 3.84 (1H, s, OH), 4.02-4.05 (1H, m), 4.40 (1H, m); $^{13}\mathrm{C}$ NMR (100 MHz, CDCl₃) δ -5.5 (2C ×3/4), -5.4 (2C ×1/4), -4.8 (1C ×1/4), $-4.7 (1C \times 3/4), -4.4 (1C \times 1/4), -4.3 (1C \times 3/4), 12.61 (1C \times 3/4), 12.64$ $(1C \times 1/4)$, 17.7 (1C), 18.20 $(1C \times 1/4)$, 18.21 $(1C \times 3/4)$, 20.2 $(1C \times 3/4)$, 20.4 $(1C \times 1/4)$, 25.66 $(3C \times 1/4)$, 25.69 $(3C \times 3/4)$, 25.9 $(3C \times 3/4)$, 26.0 $(3C \times 1/4)$, 36.5 (1C), 37.2 (1C \times 1/4), 37.5 (1C \times 3/4), 39.0 (1C \times 3/4), 50.0 (1C \times 1/4), 50.1 (1C \times 3/4), 50.2 (1C \times 1/4), 50.3 (1C \times 3/4), 51.2 (1C \times 1/4), 51.3 $(1C\times 3/4),\,53.1\,\,(1C\times 1/4),\,53.2\,\,(1C\times 3/4),\,55.9\,\,(1C\times 3/4),\,56.5\,\,(1C\times 1/4),$ 56.6 (1C \times 3/4), 59.5 (1C), 61.3 (1C \times 1/4), 61.5 (1C \times 3/4), 62.1 (1C \times 3/4), $62.4 (1C \times 1/4), 63.3 (1C \times 3/4), 63.5 (1C \times 1/4), 70.8 (1C), 73.0 (1C \times 1/4),$ 73.2 (1C \times 3/4), 75.8 (1C \times 1/4), 75.9 (1C \times 3/4), 216.8 (1C \times 3/4), 217.7 $(1C \times 3/4)$, 218.1 $(1C \times 1/4)$, 221.0 $(1C \times 1/4)$, two ¹³C peaks of the minor diastereomer were not observed; HRMS (ESI) calcd for C32H61NO6Si2Na [M+Na]⁺ 634.3930, found 634.3943.

Compound 4d

According to the general procedure B, a 3.6:1 mixture of 14c and 4d (22.3 mg, 44.9 µmol for 14c, 12.5 µmol for 4d) was obtained by using cyclopentenone 2c (enantiopure, 57.8 mg, 375 μmol), aldehyde 3b (38 μl, 370 μmol), iodide 1 (er = 9:1, 42.1 mg, 125 μ mol) and Et₃B (0.99 μ m in hexane, 380 μ l, 380 μ mol) in CH₂Cl₂ (250 μl). The crude was purified by flash column chromatography on silica gel (4 g, CH₂Cl₂/EtOAc 100:1 to 5:1). The yields were calculated to be 36% for 14c and 10% for 4d, respectively. A small amount of the mixture was repurified by flash column chromatography to obtain pure 4d for characterization. $[\alpha]_D^{25}$ 9.1 (c 0.60, CHCl₃); IR (film) ν 3491, 2974, 2934, 2812, 1736, 1705, 1455, 1382, 1243, 1207, 1156, 1114, 1042, 915 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 1.02 (3H, t, J = 7.5 Hz, NCH₂CH₃), 1.20–1.32 (3H, m), 1.35 (3H, s, acetonide), 1.45-1.49 (1H, m), 1.59 (3H, s, acetonide), 1.77 (1H, br s, OH), 2.04-2.07 (1H, m), 2.25-2.30 (2H, m), 2.33-2.40 (1H, m), 2.49 (1H, d, J = 10.3 Hz, H-9), 2.85–2.87 (2H, m), 2.86 (1H, d, J = 11.5 Hz), 3.19 (1H, d, J = 9.8 Hz, H-18a), 3.29 (3H, s, OMe), 3.31 (1H, d, J = 9.8 Hz, H-18b), 4.00 (1H, s), 4.65-4.69 (2H, m, H-12 and H-13), 4.75 (1H, d, J = 10.3 Hz, H-8), 7.26–7.37 (5H, m, aromatic); ¹³C NMR (100 MHz, CDCl₃) δ 12.3, 14.1, 18.4, 22.3, 24.1, 26.5, 37.2, 49.9, 51.0, 51.7, 56.5, 59.5, 61.7, 62.4, 76.0, 79.4, 80.2, 111.5, 127.7, 128.2, 128.3, 140.3, 216.1, 217.4; HRMS (ESI) calcd for C₂₇H₃₇NO₆Na [M+Na]⁺ 494.2513, found 494.2497.

CONFLICT OF INTEREST

The authors declare no conflict of interest.

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- Wang, F. P. & Chen, Q.-H. in *The Alkaloids: Chemistry and Biology*, Vol. 69: (ed. Cordell G. A.) 1–577 (Elsevier, New York, 2010).
- 2 Goodall, K. J., Barker, D. & Brimble, M. A. A review of advances in the synthesis of analogues of the *Delphinium* alkaloid methyllycaconitine. *Synlett* 2005, 1809–1827 (2005).
- 3 Cherney, E. C. & Baran, P. S. Terpenoid-alkaloids: their biosynthetic twist of fate and total synthesis. *Isr. J. Chem.* 51, 391–405 (2011).
- 4 Hamlin, A. M., Kisunzu, J. K. & Sarpong, R. Synthetic strategies toward hetidine and hetisine-type diterpenoid alkaloids. Org. Biomol. Chem. 12, 1846–1860 (2014).
- Liu, X.Y. & Qin, Y. Ongoing pursuit of diterpenoid alkaloids: a synthetic view. Asian J. Org. Chem. 4, 1010–1019 (2015).
- Zhu, G., Liu, R. & Liu, B. Total synthesis of atisane-type diterpenoids and related diterpenoid alkaloids. *Synthesis* 47, 2691–2708 (2015).
- Wiesner, K. The total synthesis of racemic talatisamine. *Pure Appl. Chem.* 41, 93–112 (1975).
- 8 Wiesner, K. Total synthesis of delphinine-type alkaloids by simple, fourth generation methods. *Pure Appl. Chem.* 51, 689–703 (1979).
- 9 Shi, Y., Wilmot, J. T., Nordstrøm, L. U., Tan, D. S. & Gin, D. Y. Total synthesis, relay synthesis, and structural confirmation of the C18-norditerpenoid alkaloid neofinaconitine. *J. Am. Chem. Soc.* 135, 14313–14320 (2013).
- 10 Marth, C. J. et al. Network-analysis-guided synthesis of weisaconitine D and liljestrandinine. Nature 528, 493–498 (2015).
- 11 Nishiyama, Y., Yokoshima, S. & Fukuyama, T. Total synthesis of (-)-cardiopetaline. Org. Lett. 18, 2359–2362 (2016).
- 12 Tabuchi, T., Urabe, D. & Inoue, M. Construction of the fused pentacycle of talatisamine via a combination of radical and cationic cyclizations. *J. Org. Chem.* **81**, 10204–10213 (2016).
- 13 Hagiwara, K., Tabuchi, T., Urabe, D. & Inoue, M. Expeditious synthesis of the fused hexacycle of puberuline C via a radical-based cyclization/translocation/cyclization process. *Chem. Sci.* 7, 4372–4378 (2016).
- 14 Tojino, M. & Ryu, I. in *Multicomponent Reactions* (eds Zhu J. & Bienaymé H.) 169–198 (Wiley-VCH, Weinheim, 2005).
- 15 Nicolaou, K. C., Edmonds, D. J. & Bulger, P. G. Cascade reactions in total synthesis. Angew. Chem. Int. Ed. 45, 7134–7186 (2006).
- 16 Godineau, E. & Landais, Y. Radical and radical-ionic multicomponent processes. Chem. Eur. J. 15, 3044–3055 (2009).
- 17 Pellissier, H. Stereocontrolled domino reactions. Chem. Rev. 113, 442-524 (2013).
- 18 Urabe, D., Asaba, T. & Inoue, M. Convergent strategies in total synthesis of complex terpenoids. Chem. Rev. 115, 9207–9231 (2015).
- 19 Inoue, M. Evolution of radical-based convergent strategies for total syntheses of densely oxygenated natural products. Acc. Chem. Res. 50, 460–464 (2017).
- 20 Kamimura, D., Urabe, D., Nagatomo, M. & Inoue, M. Et₃B-mediated radical-polar crossover reaction for single-step coupling of 0,Te-acetal, α,β-unsaturated ketones, and aldehydes/ketones. *Org. Lett.* 15, 5122–5125 (2013).
- 21 Kamimura, D., Nagatomo, M., Urabe, D. & Inoue, M. Expanding the scope of Et₃B/O₂-mediated coupling reactions of O,Te-acetal. *Tetrahedron* 72, 7839–7848 (2016).
- 22 Walton, J. C. Bridgehead radicals. Chem. Soc. Rev. 21, 105-112 (1992).
- 23 Kraus, G. A., Andersh, B., Su, Q. & Shi, J. Bridgehead radicals in organic chemistry. An efficient construction of the ABDE ring system of the lycoctonine alkaloids. *Tetrahedron Lett.* 34, 1741–1744 (1993).
- 24 Urabe, D., Yamaguchi, H. & Inoue, M. Application of α-alkoxy bridgehead radical for coupling of oxygenated carbocycles. Org. Lett. 13, 4778–4781 (2011).
- 25 Schnermann, M. J. & Overman, L. E. A concise synthesis of (-)-aplyviolene facilitated by a strategic tertiary radical conjugate addition. *Angew. Chem. Int. Ed.* 51, 9576–9580 (2012).
- 26 Lackner, G. L., Quasdorf, K. W. & Overman, L. E. Direct construction of quaternary carbons from tertiary alcohols via photoredox-catalyzed fragmentation of tert-alkyl N-phthalimidoyl oxalates. J. Am. Chem. Soc. 135, 15342–15345 (2013).
- 27 Rowlands, G. J. Radicals in organic synthesis. Part 1. Tetrahedron 65, 8603–8655 (2009).
- 28 Rowlands, G. J. Radicals in organic synthesis. Part 2. Tetrahedron 66, 1593–1636 (2010).
- 29 Beraud, V., Gnanou, Y., Walton, J. C. & Maillard, B. New insight into the mechanism of the reaction between α,β-unsaturated carbonyl compounds and triethylborane (Brown's reaction). *Tetrahedron Lett.* 41, 1195–1198 (2000).
- 30 Nozaki, K., Oshima, K. & Utimoto, K. Trialkylborane as an initiator and terminator of free radical reactions. Facile routes to boron enolates via α -carbonyl radicals and aldol reaction of boron enolates. *Bull. Chem. Soc. Jpn* **64**, 403–409 (1991).
- 31 Ollivier, C. & Renaud, P. Organoboranes as a source of radicals. *Chem. Rev.* 101, 3415–3434 (2001).
- 32 Godineau, E. & Landais, Y. Multicomponent radical processes: synthesis of substituted piperidinones. J. Am. Chem. Soc. 129, 12662–12663 (2007).

- 33 Kraus, G. A. & Shi, J. Reactions of bridgehead halides. A synthesis of modhephene, isomodhephene, and *epi*-modhephene. *J. Org. Chem.* **56**, 4147–4151 (1991).
- 34 Ihara, M., Suzuki, M., Fukumoto, K. & Kabuto, C. Asymmetric total synthesis of atisine via intramolecular double Michael reaction. *J. Am. Chem. Soc.* **112**, 1164–1171 (1990).
- 35 Hudlicky, T. & Reed, J. W. Applications of biotransformations and biocatalysis to complexity generation in organic syntheses. *Chem. Soc. Rev.* **38**, 3117–3132 (2009).
- 36 Akita, H. Recent advances in the synthesis of biologically active natural products using biocatalyst. *Heterocycles* **78**, 1667–1713 (2009).
- 37 Sasano, Y. et al. Highly chemoselective aerobic oxidation of amino alcohols into amino carbonyl compounds. Angew. Chem., Int. Ed. 53, 3236–3240 (2014).
- 38 Bickley, J. F., Roberts, S. M., Santoro, M. G. & Snape, T. J. Synthesis and revision of the stereochemistry of a cyclopentenone natural product isolated from ascomycete strain A23-98. *Tetrahedron* **60**, 2569–2576 (2004).
- 39 Inoue, T. & Mukaiyama, T. Regio- and stereoselective cross-aldol reactions via dialkylboryl triflates. Bull. Chem. Soc. Jpn 53, 174–178 (1980).
- Evans, D. A., Nelson, J. V., Vogel, E. & Taber, T. R. Stereoselective aldol condensations via boron enolates. *J. Am. Chem. Soc.* 103, 3099–3111 (1981).

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