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Assessing the feasibility of near-ambient conditions superconductivity in the Lu-N-H system

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The report of near-ambient superconductivity in nitrogen-doped lutetium hydrides (Lu-N-H) has generated a great interest. However, conflicting results raised doubts regarding superconductivity. Here, we combine high-throughput crystal structure predictions with a fast predictor of superconducting critical temperature (T_c) based on electron localization function to shed light on the properties of Lu-N-H at 1 GPa. None of the predicted structures supports high-temperature superconductivity and the inclusion of nitrogen in the crystal structure predictions leads to more insulating structures than metallic ones in quantity. Despite the lack of near-ambient superconductivity, we consider alternative metastable templates and study their T_c and dynamical stability including quantum anharmonic effects. Lu₄H₁₁N exhibits a T_c of 100 K at only 20 GPa, a large increase compared to 30 K of its parent LuH₃. Interestingly, it has a similar X-ray pattern to the experimental one. The LaH₁₀-like LuH₁₀ and CaH₆-like LuH₆ become high-temperature superconductors at 175 GPa and 100 GPa, with T_c of 286 K and 246 K, respectively. Our findings suggest that high-temperature superconductivity is not possible in stable phases at near-ambient pressure. However, at a slightly enhanced pressure of 20 GPa, high- T_c superconductivity emerges in Lu-H-N, and metastable room-temperature superconducting templates persist at high pressures.

Superconductivity is one of the most fascinating physical properties of matter. Ever since its discovery in mercury below 4.2 K in 1911, humanity has embarked on a restless quest for room-temperature superconductivity at ambient conditions, ceaselessly moving the field forward. Ashcroft suggested that the chemical precompression exerted by the host atoms could boost the superconducting critical temperatures (T_c) of hydrogen-rich compounds at lower pressures than pure metallic hydrogen¹. This idea was further propelled by ab initio crystal structure prediction techniques at high pressure, which could predict thermodynamically stable high-T_c crystal structures²⁻⁸. The main breakthrough arrived in 2015 when Drozdov et al.⁹ observed superconductivity at 203 K in H₃S at 155 GPa, a compound that had been anticipated theoretically by Duan et al.¹⁰. Since then, experiments have reported superconductivity above 200 K in several binary systems such as LaH₁₀ (~250 K at 150 GPa^{11,12}), CaH₆ (~215 K at 172 GPa¹³), YH₉ (~243 K at 201 GPa¹⁴), and YH₆ (~224 K at 166 GPa^{14,15}). Remarkably, ab initio calculations support or have even anticipated all these discoveries^{7,8,10,15-18}.

Considering that standard ab initio crystal structure searches of binary hydrides have been exhausted, the attention is shifting towards ternary hydrides, which offer more possibilities due to the increased complexity of the phase space^{6,19}. This recent effort has led to the prediction of ternary superhydrides with high T_c s at moderate (below 100 GPa) and even ambient pressures, at which binary superhydrides do not seem to sustain a critical temperature larger than 100 K^{20–28}. In particular, Dasenbrock-Gammon et al. have recently reported experimental evidence of superconductivity in nitrogen-doped lutetium hydride (Lu-N-H) samples with a room-temperature T_c of 294 K at nearly ambient pressure (1 GPa)²⁹.

This last claim has driven a surge of interest and excitement, but a lively discussion and polemic too. Numerous experimental and theoretical efforts have been made lately trying to replicate or explain the findings uncovered by Dasenbrock-Gammon et al. However, emerging evidence challenges the claim of room-temperature superconductivity. The impressive color change with pressure in their study is suggested to arise from $Fm\bar{3}m$ LuH₂ due to the presence of an undamped interband plasmon that enters the visible range

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with increasing pressure and, thus, does not have any impact on superconductivity³⁰, contrary to the original claim²⁹. Further experimental and theoretical studies³¹⁻³⁵ support that the color change can be explained alone by LuH₂, a compound that is known since a long time in which Lu atoms form a face-centered cubic (fcc) lattice and H atoms occupy interstitial tetrahedral sites³⁶. In addition, several experimental and theoretical investigations of the X-ray powder diffraction (XRD) point out that the major peaks in Dasenbrock-Gammon et al.'s study should mostly come from $Fm\bar{3}m$ LuH₂^{31,32,34,37-39}. These studies have indicated that the parent phase of the Lu-N-H is more likely to be $Fm\bar{3}m$ LuH₂, rather than the $Fm\bar{3}m$ LuH₃ as claimed by Dasenbrock-Gammon et al. However, all the existing experimental and theoretical studies^{30,35,39-41} have shown that LuH₂ is not superconducting or only shows theoretical T_c on the order of 0.01 K at 0~1 GPa. Ab initio crystal structure predictions exploring different stoichiometries of nitrogen-doped lutetium hydrides do not predict any phase with near-ambient superconductivity^{39,42,43}. Furthermore, Ming et al.³² and Cai et al.44 have successfully obtained nitrogen-doped lutetium hydrides, and asserted that the crystal structures of their samples were the same as those synthesized by Dasenbrock-Gammon et al.²⁹. Specifically, the lattice constants of two face-centered-cubic phases (5.03 Å and 4.755 Å) in Cai et al.'s study are in excellent agreement with Dasenbrock-Gammon et al.'s sample A (5.0289 Å) and sample B (4.7529 Å). However, despite this close agreement in lattice constants, neither of the two studies observed superconductivity, even at pressures up to 40 GPa and temperatures as low as 2 K. In addition, Ming et al. have proposed the sample is more appropriately represented as $LuH_{2\pm x}N_{y}$ rather than $LuH_{3-\delta}N_{\epsilon}$, suggesting that LuH_{2} is more likely to be the parent phase. It should be noted that the crystal structure of $Fm\bar{3}m$ LuH₃ is identical to $Fm\bar{3}m$ LuH₂ with an extra hydrogen atom located at the octahedral interstitial site³⁰.

The controversial results require a comprehensive understanding of the properties of Lu-N-H system. Herein, we report high-throughput crystal structure calculations in the Lu-H and Lu-N-H systems and screen the potential $T_{\rm c}$ of the predicted compounds with a simple descriptor based on electronic properties⁴⁵. Since LuH₂ and LuH₃ have been suggested to be the potential parent phases of the near-ambient superconducting Lu-N-H, we focus the structural search on derivations of both of them with and without nitrogen. By screening more than 15,000 structures at 1 GPa, our prediction has resulted in the discovery of 638 phases located within 0.24 eV per atom above the convex hull, a reasonable limit for the synthesizability of metastable phases, with 214 of them metallic. Our results suggest that these 1-GPa-phases are improbable to manifest high-temperature superconductivity, as deduced by predicting their T_{cs} with the networking value model⁴⁵. Our findings suggest that, when seeking to develop metallic lutetium hydrides at 1 GPa, doping with nitrogen should be avoided as its large electronegativity removes electrons from hydrogen sites and promotes insulating phases. Moreover, we identify tens of stable metallic phases that exhibit XRD features strongly resembling the experimental XRD, suggesting that many of the predicted structures have a Lu arrangement not far from the fcc lattice and that H or N atoms occupy interstitial sites. Considering that XRD is not capable of distinguishing them, one should approach XRD structural assignments in the Lu-H-N system with care. As a result of the absence of high-temperature superconductivity at 1 GPa, we study the dynamical stability and superconducting properties of high-symmetry lutetium hydrides with and without nitrogen at higher pressures in crystal structures that favor high T_c s. We find that quantum anharmonic effects foster dynamical stability at lower pressures in all cases and strongly impact the phonon spectra. Specifically, cubic $Lu_4H_{11}N$ exhibits a high T_c of 100 K at a moderate pressure of 20 GPa. Upon increasing pressure, CaH₆-like $Im\bar{3}m$ LuH₆ and LaH₁₀-like $Fm\bar{3}m$ LuH₁₀ are found to maintain high T_cs of 246 K and 289 K, respectively, at 100 GPa and 175 GPa.

Results and discussions Phase diagram

The phase diagram for the Lu-N-H system at 1 GPa is constructed by the convex hull in Fig. 1a. The circles represent thermodynamically stable

phases forming the hull and the squares show the metastable phases up to $0.24 \text{ eV} \cdot \text{atom}^{-1}$ above the convex hull (abbreviated as $H_{\text{hull}} \leq 0.24 \text{ eV} \cdot \text{atom}^{-1}$). For each stoichiometry, only the lowest-enthalpy state is shown in the phase diagram. The enthalpy calculations are performed without considering the zero-point ionic energy, i.e., considering just the Born-Oppenheimer energy. The phase diagram includes the known stable binary phases $P\bar{3}c1$ LuH₃, $Fm\bar{3}m$ LuN, $Fm\bar{3}m$ LuH₂, and multiple artificially constructed structures (e.g. Lu_2H_5) based on $Fm\bar{3}m$ LuH_2 and LuH₃ by adding/removing H atoms at tetragonal/octahedral sites. In addition, the phase diagram comprises 638 phases predicted through highthroughput crystal structure screening over around 15,000 crystal structures. Among the 638 predicted structures within this enthalpy cutoff, there are 214 metallic and 424 insulating phases.

The $P2_1/m$ LuH₂N, with structure ID in our database of 2fu_LuH2N_389, is identified as the lowest-enthalpy state among the ternary Lu-H-N compounds at 1 GPa. The relaxed structure at 1 GPa is available in "Data availability" section, and is schematically shown in the inset of Supplementary Fig. S1. Dynamical stability in the harmonic approximation of the $P2_1/m$ LuH₂N is examined by performing finite displacement DFT calculations of $2 \times 2 \times 2$ supercells. The phonon spectra and phonon density of states (DOS), as depicted in Supplementary Fig. S1, demonstrate the absence of any imaginary modes, thereby substantiating the dynamic stability of the system. Seeing that the P21/m LuH2N is found to be thermodynamically metastable near the convex hull, it is likely to be synthesized under proper experimental conditions. However, the electronic DOS of the $P2_1/m$ LuH₂N as shown in Supplementary Fig. S2 exhibits a band gap of ~2 eV, undoubtedly excluding it to be a superconducting phase. Because $P2_1/m$ LuH₂N is the lowest-enthalpy state and is dynamically stable, the phase diagram with respect to stable $P\bar{3}c1$ LuH₃ and metastable LuH₂N is explicitly shown in Fig. 1b, in which the structures up to H_{hull} of $0.3 \text{ eV} \cdot \text{atom}^{-1}$ are displayed.

Figure 1c shows the phase diagram with respect to elements, in which the binary hydrides with $H_{\text{hull}} \leq 0.8 \text{ eV} \cdot \text{atom}^{-1}$ are included. The $Fm\overline{3}m$ phase of LuH₃, which has been proposed by Dasenbrock-Gammon et al.²⁹ as the parent compound of the room-temperature superconductor, is however found to be located above the convex hull by around 82 meV·atom⁻¹. This enthalpy difference at this pressure is not expected to be overcome by ionic zero-point energy even if quantum anharmonic effects are considered¹⁸.

To identify potential hosts for superconductivity, we have focused our attention on the stability of the 214 metallic phases with $H_{hull} \le 0.24 \text{ eV} \cdot \text{atom}^{-1}$ that have been found in our high-throughput crystal structure predictions. In order to assess the phonon stability of the 214 metallic phases in the harmonic approximation, we have performed high-throughput DFT calculations over 165,000 supercells generated by the finite displacement method (see "Methods" section). These calculations identify 57 dynamically stable metallic phases including 20 Lu₄H₇, 19 Lu₄H₉, 11 Lu₄H₇N, 2 Lu₄H₉N, 2 Lu₄H₈N, 2 Lu₃H₈N, and 1 Lu₄H₁₁N. The harmonic phonon spectra of 20 Lu₄H₇ and 19 Lu₄H₉ are displayed in Supplementary Figs. S3 and S4, respectively. Additionally, the phonon spectra of the 18 ternary Lu-N-H phases are presented in Supplementary Fig. S5. The dynamical stability of these 57 phases is well evidenced by the depicted phonon spectra, which do not show imaginary phonon modes.

The ratio of the contribution of hydrogen to the total electronic density of states (DOS) at the Fermi level is widely suggested to be an important descriptor of superconductivity in superhydrides^{7,45,46}. Therefore, the hydrogen fraction of the total DOS at the Fermi level, a.k.a. H_{DOS} , are computed for the 214 metallic phases. Table 1 lists H_{DOS} , space group symbols, and the enthalpy distances above the convex hull (H_{hull}) of the 57 stable metallic phases at 1 GPa. A complete list of the 214 metallic phases regardless of the dynamical stability is available in Supplementary Table S1. As shown in Table 1, 55 out of the 57 stable metallic phases show very low H_{DOS} ranging from 0.02 to 0.05 (or 2%–5%), and only the Lu₄H₉N (ID: 1fu_Lu4H9N_136) and Lu₃H₈N (ID: 1fu_Lu3H8N_216) with the same space group P3m1 show large H_{DOS} exceeding 20%. The low H_{DOS} widely Fig. 1 | The phase diagram of the Lu-N-H system at 1 GPa. a The phase diagram of Lu-N-H with respect to elements. Only the lowest-lying enthalpy state is shown for each stoichiometry. b The phase diagram relative to stable P3c1 LuH₃ and metastable $P2_1/m$ LuH₂N, in which $H_{hull} \le 0.3$ eV·atom⁻¹. c The phase diagram of Lu-H with respect to elements, in which $H_{hull} \le 0.8$ eV·atom⁻¹. The black circles in all panels represent the thermodynamically stable phases. Square markers refer to phases located above the convex hull, with color coding their energy distance from the convex hull. The line between LuH₃ and LuH₂ in (c) is set to dash because it is not a convex hull as those in other panels.



observed in the predicted structures signals the low critical temperature or even the absence of superconductivity of the predicted phases.

To gain more insight into the possible onset of superconductivity among the metallic phases, an understanding beyond $H_{\rm DOS}$ is necessary. However, performing electron–phonon coupling calculations for all these low-symmetry systems is not feasible. As an alternative, Belli et al. have proposed that a physical quantity termed the networking value (ϕ), which is the electron localization function value that creates an isosurface spanning throughout the whole crystal, can be used to predict easily T_c . In fact, ϕ exhibits a stronger correlation with the actual T_c of hydrogen-based superconductors than any other descriptor used so far, and can be used to predict T_c with an accuracy of about 60 K with the formula $T_c = (750\phi H_f H_{\rm DOS}^{-3} - 85)$ K, where H_f is the hydrogen fraction in the compound⁴⁵. The results of ϕ and the estimated T_c of the dynamically stable phases and all the 214 metallic states irrespective of the stability are included in Table 1 and Supplementary Table S1, respectively (an Excel file is also available in "Data availability" section). Our results indicate that the upper limit of the predicted T_c among the 214 metallic states is only 13.94 ± 60 K. These results collectively indicate that the metallic phases predicted from the high-throughput crystal structure prediction are unlikely to host high-temperature superconductivity at 1 GPa.

The interatomic distances and Bader charge analysis

In Fig. 2a, we show the shortest N-H and H-H distances for the 638 Lu-H and Lu-N-H compounds within an enthalpy of 0.24 eV-atom⁻¹ above the convex hull (i.e. $H_{\text{hull}} \leq 0.24 \text{ eV} \cdot \text{atom}^{-1}$) irrespective of their dynamical stability. For the 124 binary Lu-H systems where N is absent, the shortest

Table 1 | The 57 dynamically stable metallic states at 1 GPa

ID	<i>H</i> _{hull} (eV⋅atom ⁻¹)	Space group	H _{DOS}	ф	7 _с (±60 К)	XRD similarity (%)
1fu_Lu3H8N_216	0.24	P3m1	0.26	0.31	13.94	75.03
1fu_Lu4H7_495	0.17	R3m	0.04	0.42	-	64.48
2fu_Lu4H7_242	0.17	Cmc2 ₁	0.04	0.40	_	12.78
2fu_Lu4H9_373	0.05	Pba2	0.03	0.39	-	93.61
2fu_Lu4H9_34	0.02	Ст	0.04	0.33	-	98.53
1fu_Lu4H9_213	0.01	Cmmm	0.04	0.34	-	99.07
2fu_Lu4H9_231	0.01	P2 ₁ /m	0.04	0.34	-	99.48
1fu_Lu4H9_190	0.11	R3m	0.05	0.30	_	62.05
2fu_Lu4H9_12	0.15	Ст	0.03	0.32	-	94.71
2fu_Lu4H7_51	0.19	P4/mmm	0.02	0.43	-	68.09
2fu_Lu4H9_233	0.01	Cmmm	0.04	0.30	-	98.97
1fu_Lu4H9_139	0.06	Pmmm	0.03	0.32	-	98.66
1fu_Lu4H9_175	0.01	Pm3m	0.03	0.32	_	99.12
2fu_Lu4H7_391	0.04	Pc	0.03	0.35	-	97.01
2fu_Lu4H7_62	0.08	Fmm2	0.03	0.35	_	98.19
1fu_Lu4H9_182	0.01	Pmmm	0.03	0.31	_	98.91
2fu_Lu4H7_149	0.09	P21/m	0.03	0.33	_	98.41
2fu_Lu4H7_66	0.03	Стст	0.03	0.34	-	97.52
1fu_Lu4H7_318	0.12	Amm2	0.03	0.34	_	97.97
1fu_Lu4H7_478	0.04	Amm2	0.03	0.34	-	99.15
1fu_Lu4H7_417	0.03	P43m	0.02	0.35	_	97.87
2fu_Lu4H7_375	0.09	Cmc2 ₁	0.03	0.33	-	95.62
1fu_Lu4H7_64	0.07	Ст	0.03	0.31	-	97.45
1fu_Lu4H7_288	0.16	Amm2	0.02	0.35	-	65.62
1fu_Lu4H7_467	0.23	Pām2	0.02	0.39	_	59.39
2fu_Lu4H9_401	0.02	P4/mmm	0.04	0.26	_	99.12
2fu_Lu4H7_63	0.17	Cmc2 ₁	0.03	0.32	-	62.64
1fu_Lu4H7_29	0.16	<i>P</i> 1	0.02	0.33	-	68.36
1fu_Lu4H7N_358	0.10	R3m	0.02	0.36	_	68.35
2fu_Lu4H9_225	0.20	<i>P</i> 1	0.03	0.26	-	97.85
2fu_Lu4H9_184	0.10	I4mm	0.02	0.28	_	22.18
1fu_Lu4H7_367	0.19	<i>P</i> 1	0.03	0.28	_	48.26
1fu_Lu4H7_81	0.17	Ст	0.02	0.31	_	57.65
1fu_Lu4H9_167	0.01	I4/mmm	0.03	0.25	-	98.97
1fu_Lu4H8N_5	0.07	R3m	0.03	0.27	-	51.47
1fu_Lu4H7N_471	0.15	Pmm2	0.03	0.28	_	67.44
1fu_Lu4H7_44	0.19	C2	0.02	0.30	_	58.91
2fu Lu4H9 111	0.10	Сс	0.03	0.24	_	91.48
 1fu_Lu4H7N_166	0.15	<i>P</i> 1	0.04	0.26	_	27.14
1fu Lu4H9N 10	0.21	P3m1	0.04	0.24	_	37.17
 1fu_Lu4H9_6	0.16	<i>P</i> 1	0.04	0.22	_	64.86
2fu Lu3H8N 116	0.20	<i>P</i> 1	0.03	0.24	_	49.77
1fu Lu4H9N 136	0.04	P3m1	0.31	0.11	_	13.12
1fu Lu4H11N 251	0.19	Ст	0.05	0.19	_	66.42
1fu Lu4H7 378	0.14	Cmm2	0.02	0.27	_	79.79
1fu Lu4H7N 11	0.11	Cm	0.03	0.24	-	91.97
1fu Lu4H7 379	0.20	Pm	0.02	0.22	_	88.22
1fu Lu4H8N 105	0.07	Pm	0.03	0.21	_	74.46
1fu Lu4H7N 377	0.15	Ст	0.03	0.20	_	21.50
1fu u4H7N 491	0.14	P1	0.03	0.21	_	84 14
1fu Lu4H9 129	0.15	Imm?	0.03	0.17	_	14.80
			0.00			

Table 1 (continued) | The 57 dynamically stable metallic states at 1 GPa

ID	<i>H</i> _{hull} (eV⋅atom ⁻¹)	Space group	H _{DOS}	ф	Т _с (±60 К)	XRD similarity (%)
1fu_Lu4H7N_172	0.17	<i>P</i> 1	0.03	0.21	-	83.10
2fu_Lu4H9_67	0.14	P4 ₂ /nmc	0.02	0.20	_	40.45
1fu_Lu4H7N_488	0.17	Ст	0.02	0.22	-	67.76
1fu_Lu4H7N_372	0.15	P4mm	0.03	0.19	-	52.78
2fu_Lu4H9_420	0.04	Стст	0.03	0.15	-	93.12
1fu_Lu4H7N_494	0.15	Ст	0.02	0.16	-	63.68
1fu_Lu4H7N_67	0.02	Rām	0.03	0.06	-	53.46

The superconducting transition temperatures (T_c) are estimated by using the networking value model in ref. 45. If the predicted T_c is not greater than zero, the result is displayed as '-'. Only structures below 0.24 eV-atom⁻¹ above the convex hull are considered. H_{null} represents the enthalpy distance above convex hull. ID refers to the unique identifier of a structure entry in our database. H_{DOS} is the hydrogen fraction of the total density of states at the Fermi level, ϕ refers to the networking value. XRD similarity refers to the degree of similarity in the simulated X-ray diffraction (XRD) patterns between the structure being analyzed and the reference structure (i.e. $Fm\bar{3}m \text{ LuH}_2$).

Fig. 2 | The analysis of shortest N-H and H-H distances. a The comparison between minimal distances of N-H and H-H for 124 binary Lu-H systems and 514 ternary Lu-N-H systems. The dots and squares refer to insulating and metallic states, respectively. The color bar shows the enthalpy distance above the convex hull from 0 to 0.24 eV·atom⁻¹. b The number of compounds distributed with the shortest N-H distance. The inset shows the electron localization functions of 5 representative compounds. The isosurface values of 2fu Lu4H7 51, 1fu Lu4H11N 251, 1fu_LuH2N_366, 1fu_Lu4H7N_372, and 1fu_Lu2H5N_42 are set to 0.52, 0.75, 0.75, 0.52, and 0.75, respectively. The blue (orange) bars/isovalues indicate the metallic (insulating) nature.



N-H is artificially set to 0.0 Å. As we can see, all the binary hydrides are metallic and show an H-H distance from 1.75 to 2.5 Å. At the same pressure, the H-H distance in the thermodynamically stable $Fm\bar{3}m$ LuH₂, where all H atoms occupy the tetrahedral site (H_{tetrahedral}) is 2.48 Å. In this crystal, the distance between a tetrahedral and an octahedral site is 2.15 Å. This suggests that the hydrogen atoms in most of the binary predicted hydrides are close to a tetrahedral or octahedral arrangement in an fcc lattice. The majority of the 90 ternary metallic Lu-N-H compounds, 64 out of 90, also exhibit H-H distances larger than 1.75 Å. Observing all the previously documented superhydrides compiled in ref. 45, it can be seen that none of them exhibit $T_c > 50$ K when their shortest H-H distance exceeds 1.75 Å. This analysis further indicates that neither the binary nor ternary metallic phases

predicted from our high-throughput crystal structure prediction is likely to host high-temperature superconductivity.

Based on the shortest N-H distance, these 638 compounds can be classified into three distinct clusters, which are indicated by double-headed arrows in Fig. 2. Cluster 1, which only contains binary Lu-H compounds, consists of 124 phases that are all metallic. Cluster 2 comprises 352 insulating phases and 32 metallic phases, characterized by N-H distances <2 Å and ≥1 Å. Cluster 3 is composed of a total of 130 compounds whose shortest N-H distances are all larger than 2 Å, with 72 of them being insulating and 58 of them being metallic. The histogram in Fig. 2b, in which the vertical axis uses a logarithmic scale, explicitly displays the distribution of the number of compounds based on their shortest N-H distance. In contrast to cluster 1

comprising 124 binary Lu-H compounds, which exhibit exclusively metallic behavior in the absence of N, the ternary compounds in cluster 2 and cluster 3 containing N exhibit altered characteristics across a total of 514 compounds. The presence of N is found to lead to a significant proportion of 424 insulating phases, which accounts for approximately 82.5% of all ternary compounds. This notable proportion underscores N's crucial role in favoring the formation of more insulating structures than metallic structures in quantity.

In cluster 1 consisting of 124 Lu-H compounds, it is observed that all the shortest H-H distances span the range of 1.75-2.80 Å. The presence of relatively large H-H distances prevents the formation of H₂ molecules or H-H chains which typically contribute to forming insulating states^{42,47} although there are several exceptions such as MgH₄⁴⁸ and ScH₉⁴⁹. The observation of relatively large H-H distances provides a clear explanation for the exclusively metallic behavior of the binary compounds in cluster 1. However, as noted in Table 1 and Supplementary Table S1, the contribution of hydrogen to the electrons at the Fermi level is minor in most of these compounds, where Lu d states dominate at the Fermi level. For example, in the case of 2fu_Lu4H7_51 structure, the projected density of states in Supplementary Fig. S6 provides compelling evidence for the prominent contribution of Lu d orbitals at the Fermi level. Furthermore, in Fig. 2b, we have included an inset displaying the electron localization function of 2fu_Lu4H7_51 structure at an isosurface of 0.52. This inset illustrates the ionic bonding feature between Lu and H. In this binary Lu₄H₇, a distinct separation exists between the hydrogen atoms, characterized by the shortest H-H distance of 2.18 Å.

Among the 514 ternary compounds in clusters 2 and 3, cluster 2 accounts for approximately 74.7% (384 compounds) of all ternary Lu-N-H systems. The compounds in cluster 2 are distinguished by their small N-H distances and a noticeable propensity towards insulating character. It is found that 99.5% of the compounds in cluster 2, including 352 insulating entries and 30 metallic entries, feature a shortest N-H distance of approximately ~1.1 Å. This particularly short N-H distance corresponds to the strong covalent bonding between N and H, which is evidenced by the electron localization functions of metallic 1fu_LuH2N_366 and insulating 1fu_Lu4H11N_251 belonging to cluster 2 in Fig. 2b. Only two phases in this cluster have a shortest N-H distance larger than 1.1 Å, and both of them are metallic. It should be noted that the N-H bond length of NH₃ molecule is generally between 1.0 and 1.1 Å, and purely ammonia is perfectly insulating. Furthermore, by surveying all the compounds exclusively composed of N and H in Materials Project⁵⁰ that are located within 0.2 eV-atom⁻¹ above the convex hull, these N-H compounds all show insulating properties and the shortest N-H bond lengths are all in the range of 1.0~1.1 Å. Thus, our finding suggests that the presence of N strongly favors insulating phases due to the formation of strongly covalent bonds between N and H atoms. The strong covalent N-H bonds result both in NH or NH2 units in most Lu-N-H structures predicted in our work.

Compared to the small magnitude of the average shortest N-H distance in cluster 2, the shortest N-H distance of cluster 3 is more than twice the former, spanning from 2.0 to 3.05 Å. There are 72 insulators and 58 metals in cluster 3. In addition, Fig. 2a demonstrates a strong correlation between the electronic properties and the shortest H-H distance. Among the 71 compounds with shortest H-H distances larger than 1.75 Å belonging to cluster 3, 52 compounds (73.2%) are metallic and 19 compounds are insulating. Analogous to the binary Lu-H compounds in cluster 1, long H-H distances favor the presence of metallic states dominated by electrons coming from Lu. The example structure with ID 1fu_Lu4H7N_372, hosting the shortest N-H distance of 2.59 Å and the shortest H-H distance of 2.28 Å, shows ionic features of N and H ions. On the contrary, there are 53 insulators and 6 metals whose shortest H-H distances are smaller than 1.75 Å in this cluster. In particular, 52 of these 53 insulating compounds have the shortest H-H distances below 1.1 Å, indicating that the formation of H-H molecules plays a crucial role in determining their insulating character. To exemplify this, the electron localization isosurface with the value of 0.75 of the 1fu_Lu2H5N_42 structure is shown in Fig. 2b, which unambiguously displays the presence of an H-H molecule with an H-H distance of 0.78 Å. These analyses of the interatomic distances in Clusters 2 and 3 imply that metallic phases are favored when the shortest H-H distance exceeds 1.75 Å and the shortest N-H distance surpasses 2.0 Å. However, the ternary Lu-N-H compounds, unfortunately, do not satisfy the aforementioned conditions in most cases, thereby resulting in the promotion of insulating character.

To better understand the electronic properties of the 638 Lu-H and Lu-N-H compounds with $H_{\text{hull}} \leq 0.24 \text{ eV} \cdot \text{atom}^{-1}$, we have performed a Bader charge analysis for all these compounds. The average Bader charges of H and N for each compound are explicitly shown in Supplementary Fig. S7 versus the shortest N-H distance and shortest H-H distance. Among the 514 ternary compounds, the average Bader charge of N in metallic phases and insulating phases is -1.70e and -1.64e, respectively. It implies that N in metallic phases obtains a slightly greater number of electrons than in insulating cases. However, this difference is not significant, which can also be seen in panels (a) and (c) of Supplementary Fig. S7. In contrast to the small fluctuation of the average Bader charge of N, the average Bader charge of H in metallic phases is around -0.59e, which is nearly twice the value of insulating phases (-0.32e). This difference reveals that gaining more electrons at the hydrogen site is crucial for entering the metallic state. The Bader charge of hydrogen versus the shortest N-H distance, as exemplified in panel (b) in Supplementary Fig. S7, demonstrates that longer N-H distances can be advantageous for hydrogen atoms to gain additional electrons. In other words, the presence of nitrogen does not enhance the ability of hydrogen to acquire electrons. Thus, it is comprehensible why all the 124 binary Lu-H cases without N, in which the average Bader charge of H is -0.71e, manifest metallic states. Therefore, the charge around H atoms plays a more significant role than around nitrogen in determining electronic properties. Panels (b) and (d) in Supplementary Fig. S7 demonstrate that, among all metallic compounds, a greater acquisition of electrons occurs primarily in compounds where the shortest N-H distance is around 2.5 Å, while simultaneously ensuring that the shortest H-H distance is greater than 1.75 Å. This further suggests that, when aiming to develop metallic states in lutetium hydrides, it is advisable to avoid or keep nitrogen away from the hydrogen site. This allows H to acquire more electrons from Lu, promoting the formation of metallic states. Based on statistical data on all previously reported superconducting hydrides in ref. 45, it is found that all superconducting critical temperatures in the literature are lower than 50 K if their shortest H-H distances ≥1.75 Å. In our case, by examining all 214 metallic states, 188 of them show the shortest H-H distances ≥1.75 Å. This also implies that it is unlikely to find high-temperature superconductivity in the structures from the high-throughput structure screening at 1 GPa.

XRD comparison at 1 GPa

Among the binary lutetium hydrides that have been studied experimentally, $Fm\bar{3}m$ LuH₂ has been suggested by Xie et al.⁵¹ and Ming et al.³² to have the most similar XRD pattern as the one measured by Dasenbrock-Gammon et al.²⁹. We thus compare the XRD of LuH₂ with all 638 predicted phases from our high-throughput screening with $H_{\text{hull}} \leq 0.24 \text{ eV} \cdot \text{atom}^{-1}$ at 1 GPa. In order to compare the XRDs of different structures quantitatively, the similarity between two XRDs is computed according to the correlation function implemented in PyXtal⁵². For simpler comparison, the XRD of $Fm\bar{3}m$ LuH₂ is set as the reference and its simulated XRD is shown in Fig. 3 as a comparison with the experimental XRD reported by Dasenbrock-Gammon et al.²⁹. The XRD similarity percentages (%) for the 57 dynamically stable metallic structures are shown in Table 1, while a comprehensive list of the 214 metallic states, irrespective of their dynamical stability, is available in Supplementary Table S1. By analyzing the 214 metallic phases, we find that 48 of them show strong XRD similarity percentages larger than 90%. In contrast to the metallic phases, we do not find any structure among the 424 insulating phases showing XRD similarity percentages ≥90%. Out of these 48 metallic phases showing strong similarity in XRD, 24 of them are dynamically stable, including 9 Lu₄H₇ (mean XRD similarity ~97.68%), 14 Lu₄H₉ (mean XRD similarity ~97.26%), and 1 Lu₄H₇N₁₁ (XRD similarity ~91.97%). In Fig. 3, the dynamically stable binary hydrides with XRD



Fig. 3 | **XRD analysis and crystal structures.** The simulated XRD of predicted structures are compared with the reference state $Fm\bar{3}m$ LuH₂ and the experimental XRD in ref. 29. Structure ID, space group, and XRD similarity with respect to $Fm\bar{3}m$ LuH₂ are displayed, along with crystal structure maps.

similarity percentages larger than 99% are presented, together with the dynamically stable $Cm \ Lu_4H_7N$ (ID: 1fu_Lu4H7N_11) displaying XRD similarity higher than 90%. In addition, the crystal structures are also shown in the inset of Fig. 3. Observing the space groups and the crystal structures in Fig. 3, although all six metallic phases show very high XRD similarity, the crystal structure varies from cubic lattice with high symmetry of $Pm\overline{3}m$ to the monoclinic lattice with low symmetry of Cm. The high XRD similarity can even occur in the triclinic structures. For instance, the 2fu_Lu4H11N_570 structure, included in Supplementary Table S1, shows a large XRD similarity of 93.23% despite its space group is P1.

The pronounced similarity in XRD can be attributed to the Lu sublattice, which does not differ significantly from the fcc one even if the symmetry reduction is considerable. In Supplementary Figs. S3-S5 which show the phonon spectra and XRD similarity of the 57 dynamically stable metallic phases, we find that 20 structures exhibiting high XRD similarity greater than 95% share a common feature that their distribution of phonons in frequencies resemble that of $Fm\bar{3}m$ LuH₂ at 1 GPa displayed in Supplementary Fig. S8a. The specific characteristic is that most phonon branches are separated into two distinct frequency regions, in which some phonons are distributed below 6 THz, which have mainly a Lu character, and other branches are located around 32 THz. By comparing these spectra with the one of LuH₂ displayed in Supplementary Fig. S8b, we find that all frequencies around 32 THz in these compounds are associated to vibrations of H atoms around tetrahedral interstitial sites. This means that, despite the notable variations in space groups among the 20 metallic states in Table 1, the high XRD similarity and shared characteristics in the phonon spectra imply that these structures are similar. Therefore, we have performed an indepth analysis of the occupation of H atoms and the configuration of Lu sublattices of the 20 metallic structures. Supplementary Fig. S9 illustrates the number of H atoms that are situated at tetrahedral and octahedral sites in different configurations of Lu sublattice. In binary Lu₄H₇ compounds, all

hydrogen atoms occupy tetrahedral sites as expected, and the Lu sublattice forms the fcc structure except for 1fu_Lu4H7_64, exhibiting a bodycentered tetragonal sublattice displayed in Supplementary Fig. S10a. When it comes to Lu₄H₉, there are always eight H atoms located at tetrahedral sites and one H atom positioned at an octahedral site. The fcc Lu sublattice is also observed in these Lu₄H₉ compounds although 2fu_Lu4H9_225 has a defective fcc Lu sublattice (see Supplementary Fig. S10b) that is significantly distorted compared to the fcc Lu sublattice in $Fm\bar{3}m$ LuH₂. Our analysis suggests that the majority of metallic stable phases with high XRD similarity are derived from the cubic LuH₂, in which interstitial tetrahedral hydrogen atoms can be widely observed in fcc Lu sublattices.

Although Dasenbrock-Gammon et al.²⁹ claimed their samples were likely composed of $Fm\bar{3}m$ and Immm phases based on their XRD measurements, our analysis of XRD similarity and the arrangements of Lu sublattices suggest that XRD technique might pose challenges in identifying the crystal structure of lutetium hydrides. As evidenced above, we have identified tens of dynamically stable metallic phases that possess similar XRD features, even though some of them have a considerably lower symmetry. It is important to note that the validity of this argument is contingent on the parameters utilized in the XRD similarity calculation, particularly the parameter governing the maximum permissible time delay in the crosscorrelation function.

Potential hosts for high-temperature superconductivity above 1 GPa

Failing to find any candidate with $H_{\text{hull}} < 0.24 \text{ eV} \cdot \text{atom}^{-1}$ that can host hightemperature superconductivity in the Lu-H-N system at 1 GPa, we study high-symmetry crystal structures with potential high electron-phonon interaction that may be metastable even if farther above the convex hull: $Pm\bar{3}m$ Lu₄H₁₁N, $Im\bar{3}m$ LuH₆, and $Fm\bar{3}m$ LuH₁₀. As shown in Fig. 1b, $Pm\bar{3}m$ Lu₄H₁₁N is located 0.27 eV·atom⁻¹ above the convex hull (structure ID: 1fu_Lu4H11N_137) at 1 GPa. This structure can be derived from LuH₃ by generating the conventional cell of $Fm\bar{3}m$ LuH₃ and substituting one of the octahedral H atoms with N. In the harmonic approximation, this $Lu_4H_{11}N$, whose space group is $Pm\bar{3}m$, is unstable at 1 GPa. The other two high-symmetry binary structures considered are LuH₆ and LuH₁₀, which are artificially constructed based on the already known high-temperature hydrogen-based superconductors $Im\bar{3}m$ CaH₆¹³ and $Fm\bar{3}m$ LaH₁₀¹⁸. It is noted that the crystal structure of $Im\bar{3}m$ LuH₆ has been theoretically reported in ref. 53. At 1 GPa, the phase diagram in Fig. 1c shows that $Im\bar{3}m$ LuH₆ and $Fm\bar{3}m$ LuH₁₀ are located around 0.64 and 0.77 eV·atom⁻¹ above the convex hull, implying a highly unstable nature at 1 GPa and 0 K. They are also both dynamically unstable at this pressure at the harmonic level. The zero-point energy is not enough to make any of these structures energetically competitive at 1 GPa.

In order to investigate the impact of quantum anharmonic effects on the dynamical stability of these high-symmetry structures, we relax them within the stochastic self-consistent harmonic approximation (SSCHA)⁵⁴⁻⁵⁷ at 300 K at different pressures. This completes the prior study performed on the dynamical stability of the potential parent $Fm\bar{3}m$ LuH₂ and $Fm\bar{3}m$ LuH₃ phases³⁰. In order to assess the dynamical stability of these phases we calculate the phonons derived from the Hessian of the SSCHA free energy and check for the presence of imaginary phonon modes⁵⁶.

Our SSCHA analysis in Fig. 4a shows that $Pm\bar{3}m \operatorname{Lu}_4H_{11}N$ becomes dynamically stable at around 20 GPa and 300 K, which is comparable to the stability range of LuH₃ (6 GPa and 300 K)^{30,58}. The phonon band structure shows five distinctive regions: Lu-dominated modes below 5 THz, N-dominated modes between 8 and 11 THz, and three regions of H-dominated modes above 11 THz. The highest phonon frequency of this system (43.7 THz) is slightly larger than in the case of LuH₃ (40 THz at 20 GPa and 300 K). Using electron–phonon coupling calculations based on the crystal structure of Lu₄H₁₁N from SSCHA, the Eliashberg spectral functions, and cumulative electron–phonon coupling constant were computed and shown in Fig. 5a. Additionally, utilizing the isotropic Migdal-Eliashberg equations, we estimate the superconducting critical temperature 20 GPa

а

60

-15

Frequency (THz)

DOS

HIPN



XIR M

DOS

spectra calculated by DFPT and SSCHA, respectively. The absence of imaginary modes in the SSCHA phonon spectra indicates dynamical stability in all cases.

XUIK

Article

DOS



in this material to be 100 K at 20 GPa. This is a significant improvement compared to the potential parent compound LuH₃ at the same pressure, $T_{\rm c}$ = 30 K. The element projected Eliashberg spectral functions in Supplementary Fig. S11 shows that most of the electron-phonon coupling comes from H-dominated modes, with very small contributions from Lu and N. The major effect of N doping in this system thus comes from breaking the symmetry of tetrahedral sites occupied by H, leading to a drift of H atoms away from these positions. This in turn has a large impact on the phonon frequencies and the value of the electronic density of states at the Fermi level in this system and consequently the electron-phonon coupling. In Supplementary Table S2, the crystal structure parameters and the Wyckoff positions of the atoms of $Pm\bar{3}m$ Lu₄H₁₁N are compared with those of $Fm\bar{3}m$ LuH₃. It is clear that the H atoms, which are originally located at the highsymmetry Wyckoff sites (0.25, 0.25, 0.25) in LuH₃ are shifted to lower symmetry Wyckoff sites (0.23835, 0.23835, 0.23835). In Supplementary Fig. S12, off-centering displacements of the hydrogen atoms at the tetrahedral sites are schematically shown to demonstrate the effect of N. Because the nitrogen atom replaces the hydrogen atom located at the octahedral site, the larger atomic radius pushes the hydrogen atoms at the tetrahedral sites outwards.

The binary compound $Im\bar{3}m$ LuH₆ has been predicted as a hightemperature superconductor at 100 GPa without the inclusion of ionic quantum and anharmonic effects⁵³. We fully relax this compound at different pressures within the SSCHA to determine the phase diagram. Although we find it unstable at 100 GPa in the harmonic approximation, it is stabilized with anharmonic and quantum effects at 300 K, which is evidenced by the phonon spectra in Fig. 4b. The instability in the harmonic approximation is localized at a singular **q** point, which in ref. 53 is also showing significant softening. The **q** point that shows significant softening in ref. 53 appears in an $8 \times 8 \times 8$ grid, which could explain why the instability is not present in the previous study that uses instead a $6 \times 6 \times 6$ grid. The calculated electron–phonon coupling constant is huge, which explains the softening of this phonon mode. Due to the large electron–phonon coupling constants (see Fig. 5b), we find nearly room-temperature superconductivity at 100 GPa with T_c of 246 K in this structure, similar to the value reported in ref. 53, where a T_c of 273 K was predicted at 100 GPa at the harmonic level.

The Fm $\bar{3}$ m LuH₁₀ structure is the one adopted by the hightemperature superconductor LaH₁₀. We find that the free energy Hessian displayed in Fig. 4c does not show imaginary frequencies above 175 GPa and 300 K indicating a metastable state. We calculate the electron–phonon coupling for this structure and found that the onset of superconductivity happens practically at room temperature, $T_c = 289$ K (~16 °C). The corresponding Eliashberg spectral function is shown in Fig. 5c. In comparison to another polymorph of LuH₁₀ with a space group *P*63/*mmc*, which can also be stabilized at 200 GPa in the harmonic approximation and has a theoretically reported T_c of 134–152 K⁵⁹, *Fm* $\bar{3}m$ LuH₁₀ but it can also be stabilized dynamically at a reduced pressure.

In order to confirm the capacity of the networking value model to predict T_c s used in the high-throughput calculations, we also estimate T_c for these high-symmetry structures with it for the SSCHA structures. We found that the T_c of $Fm\bar{3}m$ LuH₃ and $Pm\bar{3}m$ Lu₄H₁₁N at 20 GPa are 47.46 ± 65 K and 99.67 ± 65 K, respectively. Furthermore, the networking value model predicts that LuH₆ at 100 GPa and LuH₁₀ at 175 GPa are superconductors with a T_c of 296.85 ± 65 K and 389.50 ± 65 K, respectively. The estimated T_c values are very close to those from accurate ab initio electron–phonon coupling calculations, especially for LuH₃ and Lu₄H₁₁N. This implies that the networking value model can be applicable to the estimation of T_c of superhydrides, even for those that have never been reported. Consequently, it justifies the use of the networking value model as a rapid estimator of the T_c of the structures obtained in the high-throughput screening (i.e. the results in Table 1).

In order to estimate whether these high-symmetry structures may be identified by diffraction experiments, we compute the XRD similarity of $Pm\bar{3}m$ Lu₄H₁₁N, $Im\bar{3}m$ LuH₆, and $Fm\bar{3}m$ LuH₁₀ at 1 GPa with reference to the cubic LuH₂. The results show that while $Im\bar{3}m$ LuH₆ and $Fm\bar{3}m$ LuH₁₀ are very different, with a similarity of 35% and 58%, respectively, $Pm\bar{3}m$ Lu₄H₁₁N has a similarity of 95%, which remarks that it may be indistinguishable in diffraction experiments from $Fm\bar{3}m$ LuH₂ and may be consistent with the observed XRD pattern in ref. 29.

Because $Pm\bar{3}n \operatorname{Lu}_4H_{23}$ has been synthesized recently in the experiment with T_c of 71 K at 218 GPa and T_c of 65 K at 181 GPa⁶⁰, we have tried to estimate its T_c from ab inito study. However, the primitive cell of $Pm\bar{3}n$ $\operatorname{Lu}_4H_{23}$ has 54 atoms in the primitive cell (see Supplementary Fig. S13), which is not practical to perform electron–phonon coupling calculations with DFPT method of Quantum Espresso. Thus, we used the networking value model to estimate the T_c of the crystal structure in which cell parameters were fixed to those obtained from XRD at 185 GPa and the internal coordinates were fully relaxed. The T_c was calculated to be 136.93 ± 65 K which is not far from the experimentally observed values within the accuracy error.

Conclusions

In conclusion, we have performed a comprehensive study by combining a high-throughput structure screening, a rapid estimator of T_c, and electron-phonon coupling calculations including quantum anharmonic effects to explore the feasibility of near-ambient superconductivity in the Lu-N-H systems. Our study suggests that the presence of nitrogen leads to more insulating structures than metallic ones in quantity, and destabilizes the lutetium hydride systems by significantly shifting them away from the convex hull. As a result, the majority of identified dynamically stable metallic phases in our investigation are binary hydrides, which are not far from the parent $Fm\bar{3}m$ LuH₂. Furthermore, we did not observe hightemperature superconductivity in all of the studied structures at 1 GPa within a reasonable threshold for metastability. We, therefore, propose that in order to have metallic and superconducting states in lutetium hydrides it is better to avoid nitrogen doping. Despite the absence of near-ambient superconductivity, the combined effect of pressure and quantum anharmonicity kindles the hope of high-temperature superconductivity in the Lu-N-H systems by realizing a T_c of 100 K at a slightly enhanced pressure of only 20 GPa in $Pm\bar{3}m$ Lu₄NH₁₁. This structure is similar to $Fm\bar{3}m$ LuH₃, but with one out of four H atoms in octahedral sites substituted by a nitrogen atom. This cubic Lu₄NH₁₁ is a thermodynamically metastable phase at 20 GPa, and it is practically indistinguishable from the parent $Fm\bar{3}m$ LuH₂ compound in diffraction experiments. Thus, this structure or variations of it provide, if any, the only possible high-T_c structure at low pressures in the Lu-H-N system. At higher pressures, above 100 GPa, CaH₆-like LuH₆ and LaH10-like LuH10 are superconductors with critical temperatures around room temperature, but with an XRD pattern incompatible with experiments.

Methods

High-throughput crystal structure prediction

State-of-the-art crystal structure prediction methods, i.e., the evolutionary algorithm implemented in CrySPY⁶¹ and the particle swarm algorithm implemented in CALYPSO^{62,63}, were combined to predict crystal structures. We have screened over 15,000 crystal structures for the Lu-N-H system. The specific stoichiometry that we have considered during the crystal structure prediction are LuH₂N, Lu₄H₇N, Lu₄H₈N, Lu₄H₉N, Lu₄H₁₁N, Lu₂H₅N, Lu₃H₈N, Lu₄H₇, and Lu₄H₉. For the crystal structure prediction of a fixed stoichiometry, we have performed several parallel crystal structure searches with fixed composition by varying the number of formula units. The size of the unit cell structures is constrained up to 32 atoms. Crystal structure predictions were performed within first-principles density functional theory (DFT) calculations using the Vienna Ab initio Simulation Package (VASP)^{64,65}. The generalized gradient approximation within the parametrization of Perdew et al. 66 was used with a Hubbard U correction in the Dudarev's form⁶⁷ to improve the accuracy of the energies of the Lu f-states. An acceptable value of U = 5.5 eV, commonly used to account for the localized *f*-states of the lanthanide systems^{29,68}, was used. The test calculations in Supplementary Fig. S14 evidenced that the Hubbard correction only affects Lu 4*f* orbitals which lie pretty far away from the Fermi level. The plane wave energy cutoff was set to 450 eV during crystal structure predictions. The **k**-point grid is generated based on the specific structure by Pymatgen⁶⁹ with a relatively high grid density of 60 points per Å⁻³ of reciprocal cell volume. To address the electronic properties of the structures, the energy cutoff was improved to 550 eV with an improved **k**-point grid density of 80 points per Å⁻³.

To calculate the hydrogen fraction of the total DOS H_{DOS} at the Fermi level, we used Sumo⁷⁰ to extract the hydrogen DOS and the total DOS. In the high-throughput DFT calculations of the DOS, we used Gaussian smearing method implemented in VASP. To guarantee the accuracy of the DOS calculation, we used a relatively small width of the smearing of 0.05 eV together with a very high k-point grid density of 120 points per Å⁻³ obtained by the k-point generation scheme of Pymatgen. The TcESTIME code has been used to estimate the superconducting T_c based on the networking value model⁴⁵. The XRD simulation and comparison of the XRD similarity were performed by PyXtal⁵².

In the PyXtal simulations for XRD, we used the wavelength of 1.5406 Å in line with the experimental study²⁹. Additional parameters were assigned default values in PyXtal, such as FWHM = 0.1 and width = 1.0.

In the phonon calculations for the crystal structures predicted from high-throughput structure screening, the VASP DFT calculations were combined with the supercell and finite displacement methods implemented in Phonopy⁷¹. Initially, the unit cells from the crystal structure predictions were further optimized with a cutoff energy of 550 eV and a k-point grid density of 130 per ${\rm \AA}^{-3}$ of reciprocal cell volume until the energy convergence reaches 10^{-8} eV and forces of each atom were less than 10^{-3} eV·Å⁻¹. Subsequently, the optimized cells were expanded to supercells for force calculations in DFT. However, considering there were many structures to be examined, it has become infeasible to consider particularly large supercells. It is noted that atomic interactions in most cases were significantly decreased with increased interatomic distances, and thus a cutoff distance of 7.2 Å was considered to be proper in setting up the supercells. Specifically, if the lattice parameter (a, b, or c) of the unit cell was smaller than 7.2 Å, it was expanded twice, otherwise, it remained unchanged. With this constraint, the 214 metallic phases with $H_{\text{hull}} \leq 0.24 \text{ eV} \cdot \text{atom}^{-1}$ from the high-throughput crystal structure predictions resulted in more than 165,000 supercells for DFT calculations.

Electron-phonon coupling calculations

We relaxed the $Pm\bar{3}m \operatorname{Lu}_{4}H_{11}N$, $Im\bar{3}m \operatorname{Lu}_{6}$ and $Fm\bar{3}m \operatorname{Lu}_{10}$ using the stochastic self-consistent harmonic approximation method⁵⁴⁻⁵⁷ on $2 \times 2 \times 2$ supercells. The number of configurations used in the minimization of the free energy was 400 for $\operatorname{Lu}_{4}H_{11}N$, and 200 for Lu_{6} and Lu_{10} . The calculation for the free energy Hessian phonons needed to configurations the dynamical stability of final structures was performed with 5000 configurations for each structure.

To calculate superconducting critical temperature for these compounds we performed electron-phonon calculations for the structures obtained through the SSCHA minimization of total free energy using density functional perturbation theory (DFPT) method as implemented in Quantum Espresso^{72,73}. Electron-phonon coupling constants were calculated on a $4 \times 4 \times 4$ **q** point grid for Lu₄H₁₁N, and an $8 \times 8 \times 8$ grid for LuH₆ and LuH_{10} . The average of the electron-phonon matrix elements over the Fermi surface was done on $24 \times 24 \times 24$ point grid and 0.012 Ry smearing for Lu₄H₁₁N and $42 \times 42 \times 42$ k point grid and 0.008 Ry smearing for LuH₆ and LuH10. Unfortunately, large electron-phonon coupling makes the full convergence of results in the LuH₆ compound very challenging. However, the estimation of critical temperature is quite robust and does not change more than 20 K between the two highest **k**-point grids $(36^3 \text{ and } 42^3)$ and two lowest smearing values (0.008 and 0.012 Ry). The Eliashberg spectral function was calculated using phonon frequencies obtained from free energy Hessian. The solution of the isotropic Migal-Eliashberg equation was

obtained with the cutoff for Matsubara frequencies of 10 times the highest phonon frequency and the reduced Coulomb interaction of $\mu^* = 0.16$. The calculation of the electron–phonon coupling for $Fm\bar{3}m$ LuH₃ was done following the parameters previously used in ref. 30.

Data availability

The data that support all the findings of this study are available in the manuscript and in the Supplementary Information. In addition, some raw data are provided at Zenodo open data repository (https://doi.org/10.5281/ zenodo.8140540)⁷⁴. This repository contains: (1) A complete list of 214 metallic phases regardless of the dynamical stability (output-including-XRD-regardless-stability-240meV.xlsx) recording the information of ID, T_{c} , networking value, H_{DOS} , XRD similarity, space group, etc. (2) Database of all the structures in VASP format whose enthalpy distance to the convex hull is less or equal to 0.24 eV per atom. (3) Both structures and dynamical matrix files in Quantum Espresso format for the electron–phonon coupling calculations for cubic Lu₄H₁₁N, Im $\bar{3}m$ LuH₆, and Pm $\bar{3}m$ LuH₁₀.

Code availability

The high-throughput crystal structural predictions were carried out using the proprietary code VASP^{64,65}, with the combination of CALYPSO^{62,63} and CrySPY75. CALYPSO is freely distributed for academic users under the license of Copyright Protection Center of China (registration No. 2010SR028200 and classification No. 61000-7500). CrySPY (https://github. com/Tomoki-YAMASHITA/CrySPY) is released under the Massachusetts Institute of Technology (MIT) License and is open source. The phonon and electron-phonon properties were calculated by Phonopy⁷¹, Quantum ESPRESSO^{72,73}, and SSCHA⁵⁴⁻⁵⁷. Phonopy (https://github.com/atztogo/ phonopy) is openly released under the BSD-3-Clause License. The SSCHA code (https://github.com/SSCHAcode/python-sscha) is open source and is based on the GNU General Public License v3.0. Quantum ESPRESSO is an open-source suite of computational tools with GNU General Public License v2.0. The crystal structure visualization software VESTA⁷⁶ is distributed free of charge for academic users under the VESTA License (https://jp-minerals.org/vesta/en/). The visualization tools for phonon and electronic properties, Sumo (https://github.com/SMTG-UCL/ sumo) and Phtools (https://github.com/yw-fang/phtools), are both based on the MIT License. The TcESTIME code is freely available at https://www. lct.jussieu.fr/pagesperso/contrera/tcestime/index.html. The PyXtal code is freely available under the MIT License (https://github.com/qzhu2017/ PyXtal).

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Author contributions

Y.-W.F. and I.E. conceived the study and planned the research. Y.-W.F. and D.D. performed the theoretical calculations. Y.-W.F. wrote the manuscript with substantial input from D.D. and I.E. I.E. secured research grants and computational resources.

Competing interests

The authors declare no competing interests.

Additional information

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